

Meteo 300 SP 2001 Problem Set #2

$$1. a. PV^\gamma = (10^5 \text{ Pa})(1000 \text{ m}^3)^{1.4} \Rightarrow V = \left(\frac{(10^5)(1000)^{1.4}}{5 \times 10^4} \right)^{\frac{1}{1.4}} = 1640 \text{ m}^3$$

$$T/p^{(\gamma-1)/\gamma} = \frac{300}{(10^5)^{.286}} \Rightarrow T = 300 \left(\frac{5 \times 10^4}{10^5} \right)^{.286} = 246 \text{ K}$$

b. Isothermal $\Rightarrow T = \text{constant}$ $PV = \text{constant}$

$$T = 300 \text{ K @ } 500 \text{ hPa}$$

$$PV = (10^5)(10^3 \text{ m}^3) \Rightarrow V = \frac{(10^5)(10^3)}{5 \times 10^4} = 2000 \text{ m}^3$$

c. Energy must be added to keep the parcel isothermal on ascent. By the ideal gas law, since $T_{\text{isothermal}} > T_{\text{adiabatic}}$, $V_{\text{isothermal}} > V_{\text{adiabatic}}$ for the same pressure.

2. Isothermal ascent $\Rightarrow T = \text{constant} \Rightarrow PV = \text{constant}$

• Work = $\int_{p_1}^{p_2} p dV$. Can integrate over volume or pressure

$$• pV = (10^5)(10^3) = 10^8 \Rightarrow V = 10^8/p \Rightarrow dV = -10^8(dp/p^2)$$

$$• \therefore \text{Work} = -\int_{10^5}^{5 \times 10^4} p(10^8) \left(\frac{dp}{p^2} \right) = -\int_{10^5}^{5 \times 10^4} 10^8 \left(\frac{dp}{p} \right) = 10^8 \ln(p) \Big|_{5 \times 10^4}^{10^5}$$

$$\text{Work} = 10^8 \ln(10^5/5 \times 10^4) = 7 \times 10^7 \text{ Joules}$$

$$\text{or } \text{Work} = + \int_{10^3}^{2 \times 10^3} \frac{10^8}{V} dV = 10^8 \ln(V) \Big|_{10^3}^{2 \times 10^3} = 7 \times 10^7 \text{ Joules}$$

3. Adiabatic ascent. $PV^\gamma = \text{constant}$. Integrate by volume

$$p = \text{const}/V^\gamma = \text{const} \cdot V^{-\gamma}$$

$$W = \int_1^2 \text{const} V^{-\gamma} dV = \text{const} \int_1^2 \frac{1}{(1-\gamma)} d(V^{1-\gamma})$$

$$= \frac{\text{const}}{1-\gamma} V^{1-\gamma} \Big|_{1000}^{1640} = \frac{10^5 \cdot (1000)^{1.4}}{1-1.4} \cdot [(1640)^{-.4} - (1000)^{-.4}]$$

$$\text{Work} = 4.5 \times 10^7 \text{ Joules}$$

In this case, the work comes at the expense of the internal energy, since $dq = 0 \Rightarrow du = -dw$