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Chemical compositions of Longmaxi shales and implications for fault stability during the extraction and storage of fuels and energy

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ABSTRACT

Shales of the Longmaxi formation are currently the most important target zone for gas recovery in the Sichuan Basin, southwest China. Considering the frequent occurrence of hydraulic fracturing implicated induced earthquakes, it is important to understand the mineralogical controls on fault stability – across this transitional basin. We recover a full stratigraphic sequence of the Longmaxi and complete XRF and TOC analyses on these shales. Major components include SiO₂, CaO, Fe₂O₃, Al₂O₃, K₂O, MgO, TiO₂ and TOC, and a transition from Ca-to Sidominant contents from the top to the base of the stratigraphic column. This variation in chemical composition reflects a change in sea level during deposition with both biogenic and hydrothermal sediments contributing to the sedimentary record. Chemical compositions are closely related to shale fault stability with high tectosilicate and/or carbonate contents promoting the potential for both instability and thus triggered earthquakes. The highest Si- and Ca-contents in sub-section 1-1 (S₁I₁₋₁) and section 2 (S₁I₂) of the Longmaxi contribute the highest contents of tectosilicates and carbonates, with these minerals potentially enabling and promoting fault instability during hydraulic fracturing. Our results have important implications for understanding the relationship of shale composition and fault stability during hydraulic fracturing for shale gas exploitation.

1. Introduction

Developing and recovering low-carbon energy and fuels (such as shale gas) greatly reduces current dependence on fossil fuels and towards the goal of net zero carbon emissions (Bilgen and Sarıkaya, 2016; Meakin et al., 2013). The Sichuan Basin is currently the most promising area for shale gas recovery in southern China and also a potential area for carbon dioxide (CO₂) storage in depleted shale reservoirs (Lai et al., 2022; Ma and Xie, 2018; Sun et al., 2020; Xiang et al., 2022; Zhao and Yang, 2015). The total natural gas resources in the Sichuan Basin are estimated to be ~ $6.6 \times 10^{12} m^3$ (Guo et al., 2020; Ma, 2017). In 2021, the annual production of shale gas in the Fuling and Changning-Weiyuan national shale gas demonstration blocks had reached ~ $7 \times 10^9 m^3$ and ~ $11 \times 10^9 m^3$, respectively (Li, 2022; Mei et al., 2022; Zhang et al., 2022). This highly productive shale gas reservoir places China as the second largest shale gas producer in the world (Gao et al., 2021; Qiu et al., 2020; Zhao et al., 2022). The Sichuan

Basin (Fig. 1) is a relatively stable region in southern China comprising four terranes - the Chuanbei depression, Chuanzhong uplift, Chuanxinan depression and Chuandongnan depression. The basin is also surrounded by the Qinling-Dabie orogenic belt to the north, the Songpan-Ganzi platform and Longmenshan orogen to the northwest, the Sanjiang orogenic belt to the west, the Kangdian and Qianzhong uplifts to the south, and the Xiang'e'xi depression and Jiangnan-Xuefeng orogenic belt to the east.

Within this basin, and elsewhere, hydraulic fracturing is extensively employed in the recovery of shale gas/oil (Guo et al., 2014; Huang et al., 2019; King, 2010; Vengosh et al., 2014). This technique involves the massive injection of fluid into deep shale reservoirs to enhance reservoir permeability by creating hydraulic fractures (Hubbert and Willis, 1957; Osiptsov, 2017; Rubinstein and Mahani, 2015). The drilling of horizontal wells in the Sichuan Basin started in 2011 with systematic hydraulic fracturing beginning from 2014 (Lei et al., 2017, 2019a; Meng et al., 2019). According to the seismic data from the China Earthquake

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Networks Center (data.earthquake.cn), the Changning and Weiyuan national shale gas demonstration blocks all exhibit a sharp increase in the number of earthquakes since the initiation of systematic hydraulic fracturing (Figs. S1–S3 in the supporting information) (An et al., 2020; He et al., 2019; Wang et al., 2020). The Changning block is present in five counties (Fig. S1a), (Gao, Gong, Changning, Junlian and Xingwen counties) and the Weivuan block in three counties (Zigong, Rong and Weivuan counties) and including the cities of Zizhong and Neijiang (Fig. S1b). The earthquakes in each block are densely distributed over the hydraulic fracturing zone, i.e., within the center of Junlian county, eastern Gao county, southern Changning county, western Xingwen county and most of Gong county in the Changning block (Fig. S1a), and within the eastern Rong county, northern Zigong city, southern Zizhong county, western Neijiang city and most of Weiyuan county in Weiyuan block (Fig. S1b). Total earthquake counts at depths of 0-10 km in the Changning and Weiyuan blocks from January 2009 to June 2022 reached 28,182 and 32,532, respectively. The number of $M \ge 0$ earthquakes in each successive quarter (year) within the Changning block increased from <100 in January 2009–March 2011, to 100–500 in April 2011-December 2016, to 300-900 in January 2017-December 2018, to 1000-3400 in January 2019-December 2019, and then decreased to 500–1500 in January 2020–June 2022 (Fig. S2a). The trend in Weiyuan block is broadly consistent with the Changning block. In Weiyuan block, the number of $M \ge 0$ earthquakes in each quarter year increased from <100 in January 2009–June 2014, to 80–1000 in July 2014–September 2015, to 1500-3500 in October 2015-December 2020, and decreased to 300-1600 in January 2021-June 2022 (Fig. S2b). The earthquake frequency in both the Changning and Weiyuan blocks show an exponential growth with time (Fig. S3). These seismic data indicate a strong correlation between earthquake growth and the presence and growth of hydraulic fracturing during shale gas extraction (Schultz et al., 2020; Tan et al., 2020; Wang et al., 2015). In addition, a total of 39 earthquakes with magnitudes \geq 4 were detected in the Changning block during 2009–2022, with 34 of these (~90%) occurring after 2015. In comparison, only 9 earthquakes with magnitudes \geq 4 were recorded in the Weiyuan block over 2009–2022 with all of these occurring after 2016. These $M \geq$ 4 earthquakes are also concentrated within the hydraulic fracturing zone (Fig. S1).

The timing and spatial distribution of these seismic data are concurrent and collocated with shale gas stimulation as apparent in other studies (An et al., 2020, 2021; Lei et al., 2017, 2019b; Liu and Zahradník, 2020; Meng et al., 2019; Wang et al., 2020) and suggests causality. Currently, the target zone for shale gas recovery *via* hydraulic fracturing is located at the lower Silurian Longmaxi formation shales (Liang et al., 2012, 2014; Ma et al., 2022). These concurrent swarms of earthquakes highlight the importance of understanding fault instability in the Longmaxi shale, especially the potential for increasing hazard and severity. Considering that fault mineralogy has been shown to be one of the most important factors to control fault friction and stability (Tembe et al., 2010; Kohli and Zoback, 2013; Fang et al., 2018), it is very important to define the chemical compositions of the Longmaxi shales and correlate these compositions with the seismic potential during hydraulic fracturing.

Many previous experimental studies have explored the effects of mineralogy on fault stability using both simulated and powdered rock gouges at varied pressures, temperatures and humidity. We list the reported fault friction and stability investigations for both tectonic and reservoir applications in Table 1. These investigations generally focus on



Fig. 1. Tectonic map of southern China (modified from Xu et al., 2019). The numbers represent the ① Chuanbei depression, ② Chuanzhong uplift, ③ Chuanxinan depression, ③ Chuandongnan depression, ⑤ Dianqianbei depression, ⑥ Qianzhong uplift and ⑦ Qiannan depression. The dot dashed lines represent the boundaries of different tectonic structures (mainly areas of uplift and depression).

Table 1

List of experiments exploring the relationship between fault gouge mineralogy and fault stability for both tectonic and reservoir fault reactivations.

Fault gouge mineralogy	Normal stress	Humidity	Temperature	References
	(MPa)			
Fault gouge mineral	ogy for tectonic f	faults		
halite, kaolinite	2.5	water-	room	Bos and Spiers
		saturated		(2000)
smectite, illite,	5-150	room	room	Saffer and
quartz				Marone (2003)
smectite, quartz	75	water-	room	Takahashi et al.
		saturated		(2007)
quartz, kaolinite	5–50	water-	room	Crawford et al.
		saturated		(2008)
smectite-quartz,	12–59	water-	room	Ikari et al.
illite, chlorite	4.0	saturated		(2009)
quartz, illite,	40	water-	room	Tembe et al.
smectite	100	saturated	000 4000	(2010)
talc-serpentine/	100	water-	200–400°C	Moore and
quartz	00 50	saturated		Lockner (2011)
clay, taic and	20, 50	room	room	Teser et al.
biotito	200	water	20 60000	(2012) Lu and Ha
Diotite	200	water-	20-000 C	(2014)
tale calcite	5-50	water-	room	Giorgetti et al
taic, calcite	3-30	saturated	100111	(2015)
marble mostly	1.0_100	water-	room	(2013) Carpenter et al
calcite	1.0-100	saturated	100111	(2016)
chlorite	10	water-	room	Fagereng and
epidote.	10	saturated	100111	Ikari (2020)
amphibole				
Fault gouge mineral	ogy for reservoir	faults		
caprock,	35	dry/water-	$\sim 115^{\circ}C$	Samuelson and
reservoir		saturated		Spiers (2012)
rocks				
reservoir shales	10-30	room	room	Kohli and
				Zoback (2013)
geothermal	15, 45	water-	room	Fang et al.
reservoir		saturated		(2016)
rocks				
reservoir shales	10–55	water-	30–300	An et al. (2020)
		saturated		
reservoir granite	135–180	water-	150	Zhang et al.
		saturated		(2022)

single or paired groups of phyllosilicates, tectosilicates and carbonates. Phyllosilicates, primarily including smectite, illite, kaolinite, chlorite, mica and talc, are the most important components that control fault evolution and impart weakness to both tectonic and reservoir faults (Saffer and Marone, 2003; Crawford et al., 2008; Tesei et al., 2012; Kohli and Zoback, 2013; Lu and He, 2014; An et al., 2020). Phyllosilicate-rich fault gouges generally exhibit low frictional strength and velocity-strengthening response and promote aseismic sliding of faults. Conversely, tectosilicate-rich gouges exhibit high frictional strength and velocity-weakening behavior, promote unstable sliding and even dynamic stick-slip behaviors (Takahashi et al., 2007; Fang et al., 2016; L. Zhang et al., 2022). Carbonates are commonly found in crustal faults or reservoir caprock faults, and also show high frictional strength but exhibit complex frictional stability with response to the variations in applied effective stresses, temperatures and velocities (Giorgetti et al., 2015; Carpenter et al., 2016; An et al., 2020). Compared with simulated fault gouge as mixtures of different minerals, the frictional stability of powdered natural rocks can better represent field conditions as the fault gouge is the product of broken and crushed protolith (Vrolijk and van der Pluijm, 1999). From the literature (Kohli and Zoback, 2013; Fang et al., 2016, 2018; An et al., 2020), simulated gouges comprising powdered reservoir shales can generally be idealized as mixtures of these three primary mineral groups viz., phyllosilicates, tectosilicates and carbonates. Thus, the frictional and stability properties of powdered Longmaxi shale may be examined through the prism of these three principal mineral components.

In sum, we aim to explore links between the chemical compositions of the Longmaxi shales with shale fault instability. We realize that only analyzing the compositions of the collected shales may not totally reflect true fault stability mechanisms, as fault stability may also be controlled by many other factors - including applied temperatures, pressures and fluid chemistry. However, for mature faults of sufficient thickness fault gouges, the gouge mineralogy is a primary control on fault stability and the effects of temperature, pressure and fluid chemistry may be explored by changing the composition or physical state of the gouge zone. Therefore, exploring the impacts of shale composition and mineralogy is a straightforward and direct way to evaluate fault stability (Vrolijk and van der Pluijm, 1999; Fang et al., 2018; Zhang et al., 2019). We first recover and document a full stratigraphic sequence of the Longmaxi shales (hydrocarbon source rocks) from an open quarry within Yanjin county, Yunnan province. Then, we characterize chemical composition of the full sequence, correlate this with depositional environment and then define the link between composition, stratigraphic location and the potential for unstable and seismic fault slip and the generation of earthquakes.

2. Experimental methods

2.1. Longmaxi shale collection

A full stratigraphic sequence of the black marine Longmaxi formation shales (Nie et al., 2021; Wang et al., 2016) (Fig. 2) was recovered from the Lijiawan quarry, Yanjin county, southern Sichuan Basin at coordinates of 104°26'33.9'E and 28°07'55.4'N (denoted by the red circle in Fig. 3). A simplified stratigraphic column of the formation is shown in Fig. 3. The Lijiawan quarry is also located within the Changning block with the location also marked in Fig. S1. It can be observed that many small earthquakes also occurred close to the Lijiawan quarry. However, as most of the hydraulic fracturing operations were remote from this region, no medium or large earthquakes ($M \ge 4$) are observed. We collected 18 shale samples from the Longmaxi section at a depth of $\sim 3 m$ from the outcrop surface, to minimize the influences from the physical weathering on our shale samples (sample numbers of 1#-18#). From the stratigraphic column, the formations above and below the lower Silurian Longmaxi formation (S₁l) are lower Silurian Shiniulan (S₁s) and upper Ordovician Wufeng (O₃w) formations, respectively. Limestones dominate the Shiuniulan formation and the Wufeng formation contains mainly calcareous shale. According to the nomenclature of the marine shales in southern China (Xu et al., 2019), the Longmaxi formation (S_1) is divided into two sections, i.e., Section 1 (S₁ l_1 , represented by samples 7#-18#) and Section 2 (S_1l_2 , represented by samples1#-6#). Section 1 can be further divided into two sub-sections, i.e., Sub-Section 1-1 (S₁ l_{1-1} , represented by samples14#-18#) and Sub-Section 1-2 (S₁l₁₋₂, represented by samples 7#-13#). Section 1 (S₁l₁) comprises mainly siliceous and carbonaceous shales, with the calcareous shale dominating in Section 2 (S_1l_2). Our Longmaxi shales, recovered from the Lijiawan quarry, were all deposited in the lower Silurian period, as are the shales in the Changning and Weiyuan blocks. All were exposed in outcrop due to the multiple polycyclic tectonic movements in the Sichuan Basin (Guo, 2013; Tan et al., 2014). Consequently, the collected shale samples are representative of Changning and Weiyuan blocks.

2.2. XRF and TOC analysis of longmaxi shales

The element/oxide compositions of the collected shale samples were analyzed by X-ray fluorescence spectrometry (XRF) in the Micro Structure Analytical Laboratory of Peking University, Beijing, China. The main procedures are as follow. First, the shale samples were crushed and sieved obtain particle sizes $<74 \mu$ m, then dried at $105^{\circ}C$ for 8 h. Then, the shale powders were spread uniformly on a polyethylene sheet and loaded into a compression machine at 10 MPa pressure to form a standard sample sheet. Finally, the sample sheets were loaded into the X-ray



Fig. 2. (a) Stratigraphic section exposed in the steeply dipping beds of the outcrop Longmaxi shales in the Lijiawan quarry, southern Sichuan Basin. The arrows indicate the top and bottom of the section. (b) Outcrop of the Longmaxi shales near shale samples 8# and 9#. (c) Outcrop of the Longmaxi shales near shale sample 11#.

fluorescence spectrometer with the X-ray fluorescence intensity logged at 50 kV and 40 mA. A coarse collimator and vacuum light path were adopted during the tests with the grating diameter set to 30 mm. Quantitative analysis of element/oxide content and calibration were determined following the methods described in the standard "Analysis methods for regional geochemical sample-Part 1: Determination of aluminum oxide etc. 24 components by pressed powder pellet-X-ray fluorescence spectrometry" (Ministry of Land and Resources of China, 2016).

The total organic carbon (TOC) contents of the collected shale samples were determined from a carbon sulfur analyzer at the Laboratory Center of PetroChina Hangzhou Research Institute of Geology. The procedures are briefly described as follows. First, all shale samples were crushed and sieved to particle sizes $<20 \mu$ m with the inorganic carbon removed by hydrochloric acid solution. Subsequently, the shale powders were combusted in oxygen to convert the TOC to carbon dioxide (CO₂). Finally, the carbon dioxide content was measured by infrared detector and quantitative measurements of TOC content determined from the calibration following methods described in the standard "Determination of total organic carbon in sedimentary rock" (General Administration of Quality Supervision, Inspection and Quarantine of China, 2003).

3. Results

3.1. Chemical compositions of longmaxi shales

Results for the oxide/element contents and TOC contents are summarized in Tables A1-A2 in Appendix I. From Tables A1-A2, the major oxide contents (>1.0 *wt*%) in the Longmaxi shales include SiO₂, CaO, Fe₂O₃, Al₂O₃, K₂O, MgO and TiO₂, with trace oxides consisting of BaO, Na₂O, ZrO₂, ZnO, P₂O₅, MnO among others. Contents of SiO₂ and CaO are the highest among the oxides, with the highest SiO₂ and CaO contents approaching 65 *wt*% and 40 *wt*%, respectively. Contents of Fe₂O₃ and Al₂O₃ are primarily within 5–15 *wt*% and K₂O, MgO and TiO₂ contents are all below 10 *wt*%.

The major oxide and TOC contents of the collected Longmaxi shales relative to the sample sequence are compared in Fig. 4. It is evident that the SiO_2 content increases monotonically from <20 wt% to >60 wt% with increasing age (Fig. 4a), while the CaO content shows the opposite trend with CaO content decreasing from $\sim 40 \text{ wt\%}$ to 0 (Fig. 4b). The Fe₂O₃, Al₂O₃, K₂O and TiO₂ contents show similar trend with increasing age (Fig. 4c-f). The contents of the four oxides initially decrease within Section 2 of the Longmaxi shales (sample numbers 1#-6#), and then increase along Subsection 1-2 (sample numbers of 7#-13#), before finally decreasing along Subsection 1-1 (14#-18#), forming a "sigmoidal (S) curve". The variation of MnO content decreases with depth, similar to CaO content, decreasing from $\sim 0.15 \text{ wt\%}$ to 0 (Fig. 4g). The TOC contents in Section 2 (sample numbers of 1#-6#) and Subsection 1-2 (sample numbers of 7#-13#) are below or approaching 1 wt %, while the TOC contents increase to 2-5 wt% in Subsection 1-1 (sample numbers of 14#-18#). These results indicate that the Longmaxi shales in Subsection 1-1 have high siliceous and TOC contents, whereas the shales in Section 2 have high calcareous contents but low TOC contents. The best quality shale reservoirs are typically within TOC-rich and quartz-rich shales (Liu et al., 2020; Xu et al., 2019) and correspond to these shales in Subsection 1-1.

The relationships among SiO₂ and CaO contents with other oxide and TOC contents are compared in Figs. 5 and 6, respectively. Both CaO and MnO contents are inversely proportional to the SiO₂ content and proportional to TOC content (Fig. 5a, e and 5f). Conversely, the MnO content is proportional to the CaO content but inversely proportional to TOC content (Fig. 6d–e). The Fe₂O₃, Al₂O₃ and K₂O contents all show a bimodal distribution with an increase in SiO₂ content and the peak located at ~40 wt% SiO₂ content (Fig. 5b–d). Similarly, the Fe₂O₃, Al₂O₃



Fig. 3. Stratigraphic column of the sampled Longmaxi shales (sample numbers: 1#-18#) in the Lijiawan quarry, southern Sichuan Basin. The red solid circle indicates the sampling location of Longmaxi formation shales. Fm = Formation.

and K₂O contents also show this bimodal distribution with an increase in CaO content, but the peak is located at $\sim 10 \text{ wt}\%$ CaO content (Fig. 6a–c). In addition, the Fe₂O₃, Al₂O₃ and K₂O contents are all proportional to each other, as shown in Fig. 7.

4. Discussion

4.1. Depositional environments of longmaxi shales

The lower Silurian Longmaxi shales are primarily marine shales that are widely distributed in southwest China. From many previous studies (Boström and Peterson, 1969; Adachi et al., 1986; Yamamoto, 1987), the ratio of Al/(Al + Fe + Mn) can be regarded as an index for the hydrothermal or biogenic contribution to sediments. The ratio of Al/(Al + Fe + Mn) in sedimentary rocks from hydrothermal environments is characterized by a lowermost value of 0.01, reaching 0.6 or more for sedimentary rocks from biogenic environments. The ratio Al/(Al + Fe + Mn)in sediments increases with increasing biogenic input to the sediments. We plot the ratios of Al/(Al + Fe + Mn), Fe/(Al + Fe + Mn) and Mn/(Al + Fe + Mn) for the 18 collected shale samples. The ratios of Al/(Al + Fe + Mn), Fe/(Al + Fe + Mn) and Mn/(Al + Fe + Mn) are insensitive to location within the stratigraphic sequence and are \sim 0.4, \sim 0.6 and <0.02, respectively (Fig. 8a-c). From previous studies (Boström and Peterson, 1969; Adachi et al., 1986) and the Al-Fe-Mn ternary diagram (Fig. 8d), the Longmaxi shales are located at the transition zone between biogenic and hydrothermal sedimentary environments and these reflect that both biogenic and hydrothermal inputs contribute to the sedimentary record of the Longmaxi shales.

Southwest China experienced a crustal thickening and changes in sea level (Charvet, 2013) during the Silurian with this change resulting in variation in the sedimentary environment. According to the nomenclature for the sedimentary environments of Paleozoic marine shales in southwest China (Xu et al., 2019), the sedimentary environments of the Longmaxi formation shales underwent a transition from an intrashelf to a shallow-shelf environment between Sub-Section 1-1 to Section 2. The oceanic fauna in both intrashelf and shallow-shelf environments contributed to the growth of biogenic sediments in the Longmaxi shales. However, the intra-shelf environment was exposed to ample sunshine with thriving ocean fauna. The skeletal remains of these ocean fauna contributed to the enrichment of siliceous contents (SiO₂) and TOC in the intra-shelf environment (Hawkins et al., 2017). With a drop in sea level, the population of fauna reduced on the shallow-shelf and thus lower contents of siliceous contents and TOC are present, consistent with the oxide variation in the Longmaxi shales (Fig. 4). In addition, high calcareous contents are present in the shallow-shelf sediments (Mount, 1984; Rankey, 2004), with this is in accordance with the observed high calcium oxide content in Section 2.

In addition, the ratio of SiO_2/Al_2O_3 can also be an indicator for the sedimentary environment, thus we plot ratio SiO_2/Al_2O_3 with depth in section (increasing age) in Fig. 9a (Boström et al., 1972). Higher SiO_2/Al_2O_3 ratios imply that more biogenic sediments are deposited. Accordingly, the SiO_2/Al_2O_3 ratio approaches 2 in Sub-Section 1-2



Fig. 4. The major oxide and TOC contents for the 18 Longmaxi shales, (a) SiO₂, (b) CaO, (c) Fe₂O₃, (d) Al₂O₃, (e) K₂O, (f) TiO₂, (g) MnO, and (h) TOC.



Fig. 5. SiO₂ content versus other oxide and TOC contents: (a) CaO, (b) Fe₂O₃, (c) Al₂O₃, (d) K₂O, (e) MnO, and (f) TOC.

(sample numbers 7#-13#) and Section 2 (sample numbers 1#-6#), but increases dramatically in Sub-Section 1-1. The siliceous contents in Sub-Section 1-1 are mainly a result of the biogenic sediments - an environment beneficial for the enrichment of shale gas. Conversely, anoxic conditions are beneficial for the enrichment of TOC and can be indexed by the V/Cr and Ni/Co ratios (Jones and Manning, 1994). Ratios of V/Cr < 2 or Ni/Co < 5 represent oxygen-enriched environments, whereas the ratios V/Cr > 2 or Ni/Co > 5 indicate oxygen-lean environments. From Fig. 9b, the ratio of V/Cr is lower than 2 in Subsection 1-2 (sample numbers 7#-13#) and Section 2 (sample numbers 1#-6#), but generally higher than 2 in Sub-Section 1-1 (sample numbers 14#-18#). From Fig. 9c, the ratio of N/Co is lower than 5 in Subsection 1-2 (sample numbers 7#-13#) and Section 2 (sample numbers 1#-6#), but generally higher than 5 in Sub-Section 1-1 (sample numbers 14#-18#). These indicate that the anoxic sedimentary environments in Sub-Section 1-1 and the higher values of TOC in Sub-Section 1-1 are controlled by the anoxic sedimentary environments.

4.2. Relationship of shale fault stability and chemical composition

Induced seismicity during shale gas/oil recovery is a contemporary global concern – with the December 2018 M_L 5.7 earthquake in the Sichuan Basin regarded as the largest hydraulic fracturing-triggered

event reported worldwide (Atkinson et al., 2020; Lei et al., 2019b). Hydraulic fracturing in the Sichuan Basin is currently mainly performed in the lower Silurian Longmaxi formation - with fault stability in these shales potentially affected by various intrinsic and extrinsic factors. Intrinsic factors include fault roughness and fault gouge composition (Fryer et al., 2022; Giorgetti et al., 2015; Ikari et al., 2009; Morad et al., 2022), with the latter potentially indexed to mineralogy and deposition environment (Ingles et al., 1998; Yan et al., 2018; Xu et al., 2019; Feng et al., 2023). Extrinsic factors mainly include ambient stresses, temperatures, fluid environments and chemical effects (Moore and Lockner, 2011; Scuderi et al., 2014). For the deep mature faults present in the Longmaxi formation, fault gouge is generally formed due to the large shear offsets that have developed over geological time, with fault gouge composition controlling fault reactivation and related instability (Vrolijk and van der Pluijm, 1999). In addition, this fault gouge is the product from the cumulative wear and crushing of the protolith (Engelder, 1974). Although the role of shale composition cannot be isolated from other factors in evaluating fault stability as other external factors including pressure, temperature and chemical effects could also affect fault stability by changing gouge composition or physical state (e. g., porosity) (Samuelson et al., 2009; Moore and Lockner, 2011; Carpenter et al., 2016). In addition, for injection-induced earthquakes, the fluid injection should always be the main cause. But if the subsurface



Fig. 6. CaO content versus other oxide and TOC contents: (a) Fe₂O₃, (b) Al₂O₃, (c) K₂O, (d) MnO, and (e) TOC.



Fig. 7. Relationships among Fe₂O₃, Al₂O₃ and K₂O contents: (a) Fe₂O₃ versus Al₂O₃ contents, (b) Fe₂O₃ versus K₂O contents, and (c) Al₂O₃ versus K₂O contents.



Fig. 8. Proportions of (a) Al/(Mn + Fe + Al), (b) Fe/(Mn + Fe + Al), and (c) Mn/(Mn + Fe + Al). (d) Al-Fe-Mn ternary diagram inferring the depositional environment of the Longmaxi shales. The orange and cyan regions represent the biogenic and hydrothermal sedimentary environments, respectively.



Fig. 9. The ratios of (a) SiO_2/Al_2O_3 , (b) V/Cr and (c) Ni/Co for the collected shale samples.

faults are potentially unstable, then fluid injection could reactivate the faults and trigger the seismicity. Conversely, if the faults are potentially stable, then reactivation by fluid injection would only show aseismic creep. Hence, understanding the relationship between chemical composition and fault stability is central to defining, and potentially mitigating, the seismic hazard.

Reservoir shales typically comprise varying fractions of quartz, albite, microcline, calcite, dolomite, clay minerals (such as the smectite, illite and chlorite), and other trace minerals (such as the pyrite) (An et al., 2020; Scuderi and Collettini, 2018). From the perspective of frictional properties of minerals and hence fault instability, the minerals in reservoir shales can be indexed into three groups, comprising phyllosilicates, tectosilicates and carbonates (Fang et al., 2018). Phyllosilicates are mainly clay minerals with these minerals exhibiting low frictional strengths and promoting stable sliding behavior on reactivated faults (Vrolijk and van der Pluijm, 1999). Tectosilicates mainly comprise quartz, albite and microcline, with these minerals exhibiting high frictional strength and potentially promoting unstable sliding (Ikari et al., 2009). Carbonates include calcite and dolomite and they show similar frictional behavior to tectosilicates (Giorgetti et al., 2015). The proportion of minerals among these three mineral groups can be used as a diagnostic in determining whether a fault is approaching or moving away from instability, especially in indexing relative to the phyllosilicate content (Fang et al., 2018). However, the relative contents of these three groups of minerals are vastly different in reservoir shales from different locations. For example, the Barnett shale in the United States contains 40-60% tectosilicates, 10-40% phyllosilicates and 0-20% carbonates (Kohli and Zoback, 2013) and the Havnesville shale (USA) 15-25% tectosilicates, 25-50% phyllosilicates and 20-50% carbonates (Kohli and Zoback, 2013). Compared to these, the Longmaxi shale in the Fuling block of the Sichuan Basin returns a quartz content of 25-60% and clay content of 20-60% (Xu et al., 2019). Although a precise content in phyllosilicates required for fault instability is not defined from these various studies, a phyllosilicate content <20-30% can apparently destabilize faults in shale (Kohli and Zoback, 2013; Zhang et al., 2019). These results reflect that both or either high SiO₂ and CaO contents promote fault instability since these two oxides contribute to the predominant proportions of tectosilicates and carbonates.

Additionally, from Liang et al. (2014), Xu et al. (2019) and An et al. (2020), the minerals in the Longmaxi shales include quartz (SiO₂), albite (NaAlSi₃O₈), microcline (KAlSi₃O₈), calcite (CaCO₃), dolomite (CaMg (CO₃)₂), illite (K_{< 1}(Al, R²⁺)₂(Si, Al)Si₃O₁₀(OH)₂ \cdot nH₂O, R representing metal cations) and chlorite (Y₃(Z₄O₁₀)(OH)₂Y₃(OH)₆, Y representing metal cations, Z including Si and Al), in consistent with the tested oxide and element results (Tables A1-A2). The predominant minerals in the Longmaxi shales are quartz, calcite and illite. Although it is challenging to precisely calculate the proportions of each mineral from the oxide and element analyses (Tables A1-A2), especially for the uncertain chemical compositions in illite and chlorite, we could roughly estimate the highest contents of quartz and calcite, with the results shown in Table 2. This is also helpful in evaluating the seismic potential. The Longmaxi shales in sub-section 1-1 (S_1l_{1-1}) show the highest content of SiO₂ and this section also contains the highest potential gas contents. However, hydraulic fracturing in this section may promote fault instability due to the high content of tectosilicates. From Table 2, the highest quartz and calcite content in Longmaxi shales within section 2 (S₁l₂) are all higher than 70 wt% with some approaching 80 wt%, indicating that the illite and chlorite contents in these shales are lower than 30 to 20 wt%. The Longmaxi shales in section 2 (S₁l₂) may promote fault instability due to the higher contents of quartz (tectosilicates) and content (carbonates).

5. Conclusions

In this study, we analyzed the chemical compositions of Longmaxi shales over a full stratigraphic sequence of Longmaxi shales recovered from the Lijiawan quarry in the Sichuan Basin. The powdered shale

Table 2

The estimated quartz and calcite contents in each shale sample from Tables A1-A2. We assume that all Si and Ca contents are manifest as quartz (SiO₂) and calcite (CaCO₃), respectively, and this corresponds to the highest quartz or calcite content.

Sample No.	Section	Highest quartz (SiO ₂) content (<i>wt</i> . %)	Highest calcite (CaCO ₃) content (<i>wt</i> .%)	Highest quartz and calcite content (<i>wt</i> .%)
1#	Section 2 (S ₁ l ₂)	~18.5	~51.7	~70.2
2#		~15.2	~59.9	~75.1
3#		~16.1	~62.0	~78.1
4#		~16.4	~61.2	~77.6
5#		~14.1	~59.8	~74.2
6#		~23.5	~46.5	~70.0
7#	Subsection 1-2	~34.4	~18.7	~53.1
8#	(S_1l_{1-2}) in	~ 25.3	~38.6	~63.9
9#	Section 1 (S ₁ l ₁)	~ 31.1	~26.9	~58.0
10#		~40.5	~11.7	~52.2
11#		~43.0	~8.6	~51.6
12#		~40.7	~ 18.3	~59.0
13#		~35.3	~20.4	~55.7
14#	Subsection 1-1	~38.9	~17.0	~55.9
15#	(S_1l_{1-1}) in	~52.5	~0.9	~53.4
16#	Section 1 (S ₁ l ₁)	~51.0	~ 1.8	~52.9
17#		~62.3	~ 1.0	~63.3
18#		~64.0	~0.5	~64.5

samples were evaluated by X-ray fluorescence spectrometry (XRF) to define mineral contents and reveal the depositional environments of these shales. We use these data to define the likelihood for developing earthquakes during fault reactivation by using compositions, driven by depositional environment, as diagnostic for tectosilicate or carbonate content. The main conclusions are as follows:

- Major mineral phases in the Longmaxi shales include SiO₂, CaO, Fe₂O₃, Al₂O₃, K₂O, MgO, TiO₂ and TOC, with BaO and Na₂O and others present as trace constituents. From the top to the base of the Longmaxi formation, the chemical composition is characterized by a transition from Ca-to Si-dominated contents. The trends for oxides Fe₂O₃, Al₂O₃, K₂O, and TiO₂ from the top to the bottom of the formation are broadly the same, first decreasing and followed by an increase before again decreasing. The best-quality shales for shale gas recovery are those with the highest content of SiO₂ (brittle and potentially readily hydraulically fractured) and TOC (and hence high gas content potential) that are distributed in sub-section 1-1 (S₁l₁₋₁).
- 2. The variation in chemical compositions within the Longmaxi shale is primarily controlled by the marine sedimentary environment and through enrichment of detrital ocean fauna. The sedimentary environments of Longmaxi formation undergoes a transition from intrashelf to shallow-shelf environments with this sea level change contributing to an increase in siliceous content within the sediments. The Longmaxi shales comprise a mixture of biogenic and hydrothermal sediments resulting from these transitions in environments.
- 3. Fault instability and the potential to generate earthquakes during fault reactivation may be diagnosed relative to chemical composition of the gouges, in turn derived from the shale protolith. High SiO₂ and CaO contents contribute to higher proportions of tectosilicate or carbonate minerals and may promote fault instability during shale gas exploitation, especially in sub-section 1-1 (S_1l_{1-1}) and section 2 (S_1l_2).

CRediT authorship contribution statement

Mengke An: Conceptualization, Data curation, Formal analysis, Methodology, Writing – original draft. Fengshou Zhang: Writing – review & editing. Zhenyu Yin: Writing – review & editing. Derek Elsworth: Validation, Writing – review & editing. Rui Huang: Data curation, Validation, Writing - review & editing.

Declaration of competing interest

We declare that we have no conflict of interest.

Data availability

Data will be made available on request.

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Appendix I. Element/Oxide and TOC contents of Longmaxi shales

Table A1

Major- and trace-oxide contents (wt.%) and TOC contents (wt.%) of Longmaxi shales.

Sample No.	1#	2#	3#	4#	5#	6#	7#	8#	9#
SiO ₂	18.52 ± 0.19	15.16 ± 0.18	16.11 ± 0.18	16.42 ± 0.19	14.06 ± 0.17	23.47 ± 0.21	34.40 ± 0.24	25.26 ± 0.22	31.05 ± 0.23
CaO	33.72 ± 0.24	39.08 ± 0.24	40.47 ± 0.25	39.96 ± 0.24	39.03 ± 0.24	30.38 ± 0.23	12.18 ± 0.16	25.21 ± 0.22	17.53 ± 0.19
Fe ₂ O ₃	10.10 ± 0.15	9.69 ± 0.15	7.77 ± 0.13	8.11 ± 0.14	7.67 ± 0.13	12.24 ± 0.16	15.81 ± 0.18	12.19 ± 0.16	13.69 ± 0.17
Al ₂ O ₃	9.37 ± 0.15	7.79 ± 0.13	7.66 ± 0.13	7.38 ± 0.13	6.89 ± 0.13	9.47 ± 0.15	13.58 ± 0.17	10.11 ± 0.15	11.61 ± 0.16
K ₂ O	4.93 ± 0.11	3.98 ± 0.10	4.05 ± 0.10	3.60 ± 0.09	3.37 ± 0.09	4.91 ± 0.11	6.96 ± 0.13	5.12 ± 0.11	5.73 ± 0.12
MgO	1.67 ± 0.06	1.54 ± 0.06	1.59 ± 0.06	1.41 ± 0.06	1.34 ± 0.06	1.88 ± 0.07	$\textbf{2.73} \pm \textbf{0.08}$	2.35 ± 0.08	$\textbf{2.40} \pm \textbf{0.08}$
TiO ₂	1.03 ± 0.05	$\textbf{0.86} \pm \textbf{0.04}$	$\textbf{0.84} \pm \textbf{0.04}$	$\textbf{0.85} \pm \textbf{0.04}$	$\textbf{0.77} \pm \textbf{0.04}$	0.99 ± 0.05	1.15 ± 0.05	1.04 ± 0.05	1.06 ± 0.05
BaO	0.20 ± 0.01	0.22 ± 0.01	0.17 ± 0.01	0.16 ± 0.01	0.18 ± 0.01	0.12 ± 0.01	0.35 ± 0.02	0.30 ± 0.02	0.36 ± 0.02
Na ₂ O	0.06 ± 0.02	0.11 ± 0.02	0.08 ± 0.02	0.17 ± 0.02	0.23 ± 0.02	0.21 ± 0.02	0.27 ± 0.02	0.30 ± 0.02	0.20 ± 0.02
ZrO_2	0.15 ± 0.01	0.12 ± 0.01	0.18 ± 0.01	0.25 ± 0.01	0.13 ± 0.01	0.15 ± 0.01	0.15 ± 0.01	0.19 ± 0.01	0.15 ± 0.01
ZnO	$0.11 {\pm} {<} 0.01$	0.15 ± 0.01	$0.05 {\pm} {<} 0.01$	$0.07 {\pm} {<} 0.01$	0.19 ± 0.01	0.12 ± 0.01	$0.04{\pm}{<}0.01$	$0.03{\pm}{<}0.01$	$0.03 {\pm} {<} 0.01$
P_2O_5	$0.05 {\pm} {<} 0.01$	$0.06{\pm}{<}0.01$	$0.05 {\pm} {<} 0.01$	$0.06{\pm}{<}0.01$	$0.04{\pm}{<}0.01$	$0.06{\pm}{<}0.01$	$0.06{\pm}{<}0.01$	$0.06{\pm}{<}0.01$	$0.07 {\pm} {<} 0.01$
MnO	0.116 ± 0.006	$\textbf{0.092} \pm \textbf{0.005}$	0.112 ± 0.006	$\textbf{0.119} \pm \textbf{0.006}$	0.117 ± 0.006	0.117 ± 0.006	0.071 ± 0.004	0.114 ± 0.006	0.092 ± 0.005
V_2O_5	0.031 ± 0.003	$\textbf{0.027} \pm \textbf{0.003}$	$\textbf{0.029} \pm \textbf{0.003}$	0.033 ± 0.003	0.026 ± 0.003	$\textbf{0.040} \pm \textbf{0.003}$	$\textbf{0.044} \pm \textbf{0.002}$	$\textbf{0.040} \pm \textbf{0.003}$	0.036 ± 0.002
NiO	0.006 ± 0.002	$\textbf{0.005} \pm \textbf{0.002}$	$\textbf{0.006} \pm \textbf{0.002}$	$\textbf{0.002} \pm \textbf{0.002}$	-	$\textbf{0.012} \pm \textbf{0.002}$	$\textbf{0.016} \pm \textbf{0.002}$	$\textbf{0.007} \pm \textbf{0.002}$	0.016 ± 0.002
Cr_2O_3	0.026 ± 0.002	$\textbf{0.019} \pm \textbf{0.002}$	0.020 ± 0.002	0.013 ± 0.002	$\textbf{0.023} \pm \textbf{0.003}$	0.025 ± 0.002	0.031 ± 0.002	0.031 ± 0.002	0.031 ± 0.002
Co ₃ O ₄	0.008 ± 0.002	$\textbf{0.006} \pm \textbf{0.002}$	$\textbf{0.002} \pm \textbf{0.001}$	0.004 ± 0.002	$\textbf{0.003} \pm \textbf{0.002}$	$\textbf{0.007} \pm \textbf{0.002}$	$\textbf{0.009} \pm \textbf{0.002}$	$\textbf{0.009} \pm \textbf{0.002}$	0.007 ± 0.001
TOC	0.21	0.77	-	0.39	0.51	0.62	0.69	0.65	0.87
Sample No.	10#	11#	12#	13#	14#	15#	16#	17#	18#
SiO ₂	40.50 ± 0.25	$\textbf{43.03} \pm \textbf{0.25}$	$\textbf{40.73} \pm \textbf{0.25}$	$\textbf{35.29} \pm \textbf{0.24}$	$\textbf{38.89} \pm \textbf{0.24}$	$\textbf{52.49} \pm \textbf{0.25}$	50.95 ± 0.25	62.32 ± 0.24	64.01 ± 0.24
CaO	7.65 ± 0.13	5.61 ± 0.12	11.91 ± 0.16	13.32 ± 0.17	11.07 ± 0.16	0.57 ± 0.03	1.16 ± 0.05	0.66 ± 0.03	0.33 ± 0.02
Fe ₂ O ₃	15.64 ± 0.18	14.60 ± 0.18	11.67 ± 0.16	13.30 ± 0.17	14.15 ± 0.17	14.16 ± 0.17	15.11 ± 0.18	$\textbf{9.42} \pm \textbf{0.15}$	$\textbf{9.44} \pm \textbf{0.15}$
Al_2O_3	14.16 ± 0.17	14.91 ± 0.18	13.28 ± 0.17	13.96 ± 0.17	13.45 ± 0.17	$\textbf{12.09} \pm \textbf{0.16}$	11.62 ± 0.16	$\textbf{9.67} \pm \textbf{0.15}$	$\textbf{9.10} \pm \textbf{0.14}$
K ₂ O	$\textbf{7.14} \pm \textbf{0.13}$	$\textbf{7.19} \pm \textbf{0.13}$	$\textbf{6.03} \pm \textbf{0.12}$	$\textbf{6.60} \pm \textbf{0.12}$	$\textbf{6.16} \pm \textbf{0.12}$	$\textbf{6.05} \pm \textbf{0.12}$	$\textbf{5.95} \pm \textbf{0.12}$	5.11 ± 0.11	$\textbf{4.67} \pm \textbf{0.11}$
MgO	$\textbf{2.76} \pm \textbf{0.08}$	$\textbf{2.85} \pm \textbf{0.08}$	$\textbf{2.62} \pm \textbf{0.08}$	$\textbf{2.91} \pm \textbf{0.08}$	$\textbf{2.88} \pm \textbf{0.08}$	1.91 ± 0.07	$\textbf{2.09} \pm \textbf{0.07}$	1.51 ± 0.06	1.38 ± 0.06
TiO ₂	1.19 ± 0.05	1.24 ± 0.06	$\textbf{1.27} \pm \textbf{0.06}$	$\textbf{1.47} \pm \textbf{0.06}$	1.31 ± 0.06	1.01 ± 0.05	$\textbf{0.97} \pm \textbf{0.05}$	$\textbf{0.82} \pm \textbf{0.04}$	$\textbf{0.78} \pm \textbf{0.04}$
BaO	$\textbf{0.30} \pm \textbf{0.02}$	$\textbf{0.32} \pm \textbf{0.02}$	$\textbf{0.23} \pm \textbf{0.01}$	$\textbf{0.38} \pm \textbf{0.02}$	$\textbf{0.20} \pm \textbf{0.01}$	$\textbf{0.23} \pm \textbf{0.01}$	$\textbf{0.29} \pm \textbf{0.01}$	$0.10{\pm}{<}0.01$	$\textbf{0.18} \pm \textbf{0.01}$
Na ₂ O	0.30 ± 0.02	0.31 ± 0.02	$\textbf{0.47} \pm \textbf{0.02}$	0.50 ± 0.03	0.63 ± 0.03	0.15 ± 0.02	0.12 ± 0.02	0.22 ± 0.02	0.18 ± 0.01
ZrO ₂	0.20 ± 0.01	0.02 ± 0.01	0.07 1.0.00	0.00 1.0.01	0.41 ± 0.02	0.16 + 0.01	0.00 1.0.01	0.10 0.01	0.17 ± 0.01
	0.20 ± 0.01	0.23 ± 0.01	0.37 ± 0.02	0.28 ± 0.01	0.41 ± 0.02	0.16 ± 0.01	0.20 ± 0.01	0.13 ± 0.01	0.17 ± 0.01
ZnO	0.02 ± 0.01 $0.02 \pm < 0.01$	0.23 ± 0.01 0.28 ± 0.01	0.37 ± 0.02 0.28 ± 0.01	$0.28 \pm 0.01 \\ 0.21 \pm 0.01$	0.41 ± 0.02 0.17 ± 0.01	0.16 ± 0.01 $0.07 \pm < 0.01$	0.20 ± 0.01 $0.07 \pm < 0.01$	0.13 ± 0.01 0.26 ± 0.01	0.017 ± 0.01 $0.04 \pm < 0.01$
ZnO P ₂ O ₅	0.020 ± 0.01 $0.02\pm < 0.01$ 0.09 ± 0.01	0.23 ± 0.01 0.28 ± 0.01 $0.08 \pm < 0.01$	$\begin{array}{c} 0.37 \pm 0.02 \\ 0.28 \pm 0.01 \\ 0.09 \pm {<} 0.01 \end{array}$	$\begin{array}{c} 0.28 \pm 0.01 \\ 0.21 \pm 0.01 \\ 0.09 \pm 0.01 \end{array}$	0.41 ± 0.02 0.17 ± 0.01 0.11 ± 0.01	0.16 ± 0.01 $0.07 \pm < 0.01$ 0.16 ± 0.01	0.20 ± 0.01 $0.07 \pm < 0.01$ 0.14 ± 0.01	0.13 ± 0.01 0.26 ± 0.01 0.14 ± 0.01	0.17 ± 0.01 $0.04 \pm < 0.01$ 0.16 ± 0.01
ZnO P ₂ O ₅ MnO	$\begin{array}{c} 0.20 \pm 0.01 \\ 0.02 \pm < 0.01 \\ 0.09 \pm 0.01 \\ 0.076 \pm 0.004 \end{array}$	0.23 ± 0.01 0.28 ± 0.01 $0.08 \pm < 0.01$ 0.070 ± 0.004	0.37 ± 0.02 0.28 ± 0.01 $0.09 \pm < 0.01$ 0.077 ± 0.004	$\begin{array}{c} 0.28 \pm 0.01 \\ 0.21 \pm 0.01 \\ 0.09 \pm 0.01 \\ 0.099 \pm 0.005 \end{array}$	$\begin{array}{c} 0.41 \pm 0.02 \\ 0.17 \pm 0.01 \\ 0.11 \pm 0.01 \\ 0.066 \pm 0.003 \end{array}$	0.16 ± 0.01 $0.07 \pm < 0.01$ 0.16 ± 0.01 0.032 ± 0.002	0.20 ± 0.01 $0.07 \pm < 0.01$ 0.14 ± 0.01 0.039 ± 0.002	0.13 ± 0.01 0.26 ± 0.01 0.14 ± 0.01 0.034 ± 0.002	0.17 ± 0.01 $0.04 \pm < 0.01$ 0.16 ± 0.01 0.027 ± 0.002
ZnO P ₂ O ₅ MnO V ₂ O ₅	$\begin{array}{c} 0.02 \pm 0.01 \\ 0.02 \pm < 0.01 \\ 0.09 \pm 0.01 \\ 0.076 \pm 0.004 \\ 0.041 \pm 0.002 \end{array}$	$\begin{array}{c} 0.23 \pm 0.01 \\ 0.28 \pm 0.01 \\ 0.08 \pm < 0.01 \\ 0.070 \pm 0.004 \\ 0.043 \pm 0.003 \end{array}$	$\begin{array}{c} 0.37 \pm 0.02 \\ 0.28 \pm 0.01 \\ 0.09 \pm < 0.01 \\ 0.077 \pm 0.004 \\ 0.042 \pm 0.003 \end{array}$	$\begin{array}{c} 0.28 \pm 0.01 \\ 0.21 \pm 0.01 \\ 0.09 \pm 0.01 \\ 0.099 \pm 0.005 \\ 0.038 \pm 0.003 \end{array}$	$\begin{array}{c} 0.41 \pm 0.02 \\ 0.17 \pm 0.01 \\ 0.11 \pm 0.01 \\ 0.066 \pm 0.003 \\ 0.038 \pm 0.003 \end{array}$	$\begin{array}{c} 0.16 \pm 0.01 \\ 0.07 {\pm} {<} 0.01 \\ 0.16 \pm 0.01 \\ 0.032 \pm 0.002 \\ 0.069 \pm 0.003 \end{array}$	$\begin{array}{c} 0.20 \pm 0.01 \\ 0.07 {\pm} {<} 0.01 \\ 0.14 \pm 0.01 \\ 0.039 \pm 0.002 \\ 0.064 \pm 0.003 \end{array}$	$\begin{array}{c} 0.13 \pm 0.01 \\ 0.26 \pm 0.01 \\ 0.14 \pm 0.01 \\ 0.034 \pm 0.002 \\ 0.062 \pm 0.003 \end{array}$	$\begin{array}{c} 0.17 \pm 0.01 \\ 0.04 {\pm} {<} 0.01 \\ 0.16 \pm 0.01 \\ 0.027 \pm 0.002 \\ 0.057 \pm 0.003 \end{array}$
ZnO P ₂ O ₅ MnO V ₂ O ₅ NiO	$\begin{array}{c} 0.22 \pm 0.01 \\ 0.02 \pm < 0.01 \\ 0.09 \pm 0.01 \\ 0.076 \pm 0.004 \\ 0.041 \pm 0.002 \\ 0.018 \pm 0.002 \end{array}$	$\begin{array}{c} 0.23 \pm 0.01 \\ 0.28 \pm 0.01 \\ 0.08 \pm < 0.01 \\ 0.070 \pm 0.004 \\ 0.043 \pm 0.003 \\ 0.013 \pm 0.002 \end{array}$	$\begin{array}{c} 0.37 \pm 0.02 \\ 0.28 \pm 0.01 \\ 0.09 \pm < 0.01 \\ 0.077 \pm 0.004 \\ 0.042 \pm 0.003 \\ 0.012 \pm 0.002 \end{array}$	$\begin{array}{c} 0.28 \pm 0.01 \\ 0.21 \pm 0.01 \\ 0.09 \pm 0.01 \\ 0.099 \pm 0.005 \\ 0.038 \pm 0.003 \\ 0.008 \pm 0.002 \end{array}$	$\begin{array}{c} 0.41 \pm 0.02 \\ 0.17 \pm 0.01 \\ 0.11 \pm 0.01 \\ 0.066 \pm 0.003 \\ 0.038 \pm 0.003 \\ 0.011 \pm 0.003 \end{array}$	$\begin{array}{c} 0.16 \pm 0.01 \\ 0.07 \pm < 0.01 \\ 0.16 \pm 0.01 \\ 0.032 \pm 0.002 \\ 0.069 \pm 0.003 \\ 0.055 \pm 0.003 \end{array}$	$\begin{array}{c} 0.20 \pm 0.01 \\ 0.07 {\pm}{<} 0.01 \\ 0.14 \pm 0.01 \\ 0.039 \pm 0.002 \\ 0.064 \pm 0.003 \\ 0.063 \pm 0.003 \end{array}$	$\begin{array}{c} 0.13 \pm 0.01 \\ 0.26 \pm 0.01 \\ 0.14 \pm 0.01 \\ 0.034 \pm 0.002 \\ 0.062 \pm 0.003 \\ 0.024 \pm 0.002 \end{array}$	$\begin{array}{c} 0.17 \pm 0.01 \\ 0.04 \pm < 0.01 \\ 0.16 \pm 0.01 \\ 0.027 \pm 0.002 \\ 0.057 \pm 0.003 \\ 0.027 \pm 0.002 \end{array}$
ZnO P_2O_5 MnO V_2O_5 NiO Cr_2O_3	$\begin{array}{c} 0.20 \pm 0.01 \\ 0.02 \pm < 0.01 \\ 0.09 \pm 0.01 \\ 0.076 \pm 0.004 \\ 0.041 \pm 0.002 \\ 0.018 \pm 0.002 \\ 0.029 \pm 0.002 \end{array}$	$\begin{array}{c} 0.23 \pm 0.01 \\ 0.28 \pm 0.01 \\ 0.08 \pm < 0.01 \\ 0.070 \pm 0.004 \\ 0.043 \pm 0.003 \\ 0.013 \pm 0.002 \\ 0.029 \pm 0.002 \end{array}$	$\begin{array}{c} 0.37 \pm 0.02 \\ 0.28 \pm 0.01 \\ 0.09 \pm < 0.01 \\ 0.077 \pm 0.004 \\ 0.042 \pm 0.003 \\ 0.012 \pm 0.002 \\ 0.022 \pm 0.002 \end{array}$	$\begin{array}{c} 0.28 \pm 0.01 \\ 0.21 \pm 0.01 \\ 0.09 \pm 0.01 \\ 0.099 \pm 0.005 \\ 0.038 \pm 0.003 \\ 0.008 \pm 0.002 \\ 0.034 \pm 0.002 \end{array}$	$\begin{array}{c} 0.41 \pm 0.02 \\ 0.17 \pm 0.01 \\ 0.11 \pm 0.01 \\ 0.066 \pm 0.003 \\ 0.038 \pm 0.003 \\ 0.011 \pm 0.003 \\ 0.024 \pm 0.002 \end{array}$	$\begin{array}{c} 0.16 \pm 0.01 \\ 0.07 \pm < 0.01 \\ 0.16 \pm 0.01 \\ 0.032 \pm 0.002 \\ 0.069 \pm 0.003 \\ 0.055 \pm 0.003 \\ 0.022 \pm 0.002 \end{array}$	$\begin{array}{c} 0.20 \pm 0.01 \\ 0.07 \pm < 0.01 \\ 0.14 \pm 0.01 \\ 0.039 \pm 0.002 \\ 0.064 \pm 0.003 \\ 0.063 \pm 0.003 \\ 0.022 \pm 0.002 \end{array}$	$\begin{array}{c} 0.13 \pm 0.01 \\ 0.26 \pm 0.01 \\ 0.14 \pm 0.01 \\ 0.034 \pm 0.002 \\ 0.062 \pm 0.003 \\ 0.024 \pm 0.002 \\ 0.019 \pm 0.002 \end{array}$	$\begin{array}{c} 0.17 \pm 0.01 \\ 0.04 \pm < 0.01 \\ 0.16 \pm 0.01 \\ 0.027 \pm 0.002 \\ 0.057 \pm 0.003 \\ 0.027 \pm 0.002 \\ 0.019 \pm 0.001 \end{array}$
ZnO P_2O_5 MnO V_2O_5 NiO Cr_2O_3 Co_3O_4	$\begin{array}{c} 0.02\pm 0.01\\ 0.02\pm <0.01\\ 0.076\pm 0.004\\ 0.041\pm 0.002\\ 0.018\pm 0.002\\ 0.029\pm 0.002\\ 0.006\pm 0.002 \end{array}$	$\begin{array}{c} 0.23 \pm 0.01 \\ 0.28 \pm 0.01 \\ 0.08 \pm < 0.01 \\ 0.070 \pm 0.004 \\ 0.043 \pm 0.003 \\ 0.013 \pm 0.002 \\ 0.029 \pm 0.002 \\ 0.006 \pm 0.002 \end{array}$	$\begin{array}{c} 0.37 \pm 0.02 \\ 0.28 \pm 0.01 \\ 0.09 \pm < 0.01 \\ 0.077 \pm 0.004 \\ 0.042 \pm 0.003 \\ 0.012 \pm 0.002 \\ 0.022 \pm 0.002 \\ 0.008 \pm 0.002 \end{array}$	$\begin{array}{c} 0.28 \pm 0.01 \\ 0.21 \pm 0.01 \\ 0.09 \pm 0.01 \\ 0.099 \pm 0.005 \\ 0.038 \pm 0.003 \\ 0.008 \pm 0.002 \\ 0.034 \pm 0.002 \\ 0.005 \pm 0.002 \end{array}$	$\begin{array}{c} 0.17 \pm 0.02 \\ 0.17 \pm 0.01 \\ 0.11 \pm 0.01 \\ 0.066 \pm 0.003 \\ 0.038 \pm 0.003 \\ 0.011 \pm 0.003 \\ 0.024 \pm 0.002 \\ 0.007 \pm 0.002 \end{array}$	$\begin{array}{c} 0.16 \pm 0.01 \\ 0.07 \pm < 0.01 \\ 0.16 \pm 0.01 \\ 0.032 \pm 0.002 \\ 0.069 \pm 0.003 \\ 0.055 \pm 0.003 \\ 0.022 \pm 0.002 \\ 0.005 \pm 0.002 \end{array}$	$\begin{array}{c} 0.20 \pm 0.01 \\ 0.07 \pm < 0.01 \\ 0.14 \pm 0.01 \\ 0.039 \pm 0.002 \\ 0.064 \pm 0.003 \\ 0.063 \pm 0.003 \\ 0.022 \pm 0.002 \\ 0.010 \pm 0.001 \end{array}$	$\begin{array}{c} 0.13 \pm 0.01 \\ 0.26 \pm 0.01 \\ 0.14 \pm 0.01 \\ 0.034 \pm 0.002 \\ 0.062 \pm 0.003 \\ 0.024 \pm 0.002 \\ 0.019 \pm 0.002 \\ 0.001 \pm 0.001 \end{array}$	$\begin{array}{c} 0.07\pm0.01\\ 0.04\pm<0.01\\ 0.16\pm0.01\\ 0.027\pm0.002\\ 0.057\pm0.003\\ 0.027\pm0.002\\ 0.019\pm0.001\\ 0.006\pm0.001 \end{array}$

Table A	12
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Major- and trace-element analyses of Longmaxi shales.

Sample No.	1#	2#	3#	4#	5#	6#	7#	8#	9#
Si	8.66 ± 0.09	7.09 ± 0.08	$\textbf{7.53} \pm \textbf{0.09}$	$\textbf{7.68} \pm \textbf{0.09}$	6.58 ± 0.08	10.97 ± 0.10	16.08 ± 0.11	11.81 ± 0.10	14.51 ± 0.11
Ca	24.11 ± 0.17	$\textbf{27.94} \pm \textbf{0.17}$	$\textbf{28.94} \pm \textbf{0.18}$	28.57 ± 0.18	$\textbf{27.91} \pm \textbf{0.17}$	21.72 ± 0.16	$\textbf{8.71} \pm \textbf{0.12}$	18.03 ± 0.16	12.53 ± 0.14
Fe	$\textbf{7.07} \pm \textbf{0.11}$	$\textbf{6.78} \pm \textbf{0.10}$	5.43 ± 0.09	5.67 ± 0.10	5.36 ± 0.09	8.56 ± 0.11	11.06 ± 0.13	8.53 ± 0.11	9.57 ± 0.12
Al	$\textbf{4.96} \pm \textbf{0.08}$	$\textbf{4.12} \pm \textbf{0.07}$	$\textbf{4.05} \pm \textbf{0.07}$	3.91 ± 0.07	3.65 ± 0.07	5.01 ± 0.08	$\textbf{7.19} \pm \textbf{0.09}$	5.35 ± 0.08	$\textbf{6.15} \pm \textbf{0.08}$
K	$\textbf{4.09} \pm \textbf{0.09}$	3.31 ± 0.08	$\textbf{3.36} \pm \textbf{0.08}$	$\textbf{2.99} \pm \textbf{0.08}$	$\textbf{2.79} \pm \textbf{0.07}$	$\textbf{4.07} \pm \textbf{0.09}$	$\textbf{5.77} \pm \textbf{0.11}$	$\textbf{4.25} \pm \textbf{0.09}$	$\textbf{4.76} \pm \textbf{0.10}$
Mg	1.00 ± 0.04	$\textbf{0.93} \pm \textbf{0.04}$	$\textbf{0.96} \pm \textbf{0.04}$	$\textbf{0.85} \pm \textbf{0.04}$	$\textbf{0.81} \pm \textbf{0.04}$	1.14 ± 0.04	1.64 ± 0.05	1.42 ± 0.05	1.45 ± 0.05
Ti	$\textbf{0.62} \pm \textbf{0.03}$	$\textbf{0.52} \pm \textbf{0.03}$	0.51 ± 0.03	0.51 ± 0.03	$\textbf{0.46} \pm \textbf{0.02}$	$\textbf{0.60} \pm \textbf{0.03}$	$\textbf{0.69} \pm \textbf{0.03}$	$\textbf{0.62} \pm \textbf{0.03}$	$\textbf{0.64} \pm \textbf{0.03}$
Ba	$\textbf{0.18} \pm \textbf{0.01}$	$\textbf{0.20} \pm \textbf{0.01}$	$\textbf{0.15} \pm \textbf{0.01}$	$\textbf{0.14} \pm \textbf{0.01}$	$\textbf{0.16} \pm \textbf{0.01}$	0.11 ± 0.01	0.31 ± 0.02	$\textbf{0.27} \pm \textbf{0.01}$	0.32 ± 0.02
Na	$\textbf{0.05} \pm \textbf{0.01}$	$\textbf{0.08} \pm \textbf{0.01}$	$\textbf{0.06} \pm \textbf{0.01}$	0.13 ± 0.01	$\textbf{0.17} \pm \textbf{0.01}$	0.15 ± 0.02	$\textbf{0.20} \pm \textbf{0.01}$	$\textbf{0.22} \pm \textbf{0.01}$	0.15 ± 0.01
Zr	0.11 ± 0.01	$0.08{\pm}{<}0.01$	$\textbf{0.13} \pm \textbf{0.01}$	$\textbf{0.19} \pm \textbf{0.01}$	$0.10{\pm}{<}0.01$	0.11 ± 0.01	$\textbf{0.11} \pm \textbf{0.01}$	$\textbf{0.14} \pm \textbf{0.01}$	0.11 ± 0.01
Zn	$0.09{\pm}{<}0.01$	$0.12 {\pm} {<} 0.01$	$0.04{\pm}{<}0.01$	$0.06{\pm}{<}0.01$	$\textbf{0.15} \pm \textbf{0.01}$	$0.10{\pm}{<}0.01$	$0.03{\pm}{<}0.01$	$0.02{\pm}{<}0.01$	$0.02{\pm}{<}0.01$
Р	$0.02{\pm}{<}0.01$	$0.03{\pm}{<}0.01$	$0.02{\pm}{<}0.01$	$0.03{\pm}{<}0.01$	$0.02{\pm}{<}0.01$	$0.03{\pm}{<}0.01$	$0.03{\pm}{<}0.01$	$0.03{\pm}{<}0.01$	$0.03{\pm}{<}0.01$
Mn	$\textbf{0.090} \pm \textbf{0.005}$	0.071 ± 0.004	$\textbf{0.087} \pm \textbf{0.004}$	0.092 ± 0.005	0.091 ± 0.005	0.091 ± 0.005	0.055 ± 0.003	$\textbf{0.088} \pm \textbf{0.004}$	0.071 ± 0.004

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Table A2 (continued)

Sample No.	1#	2#	3#	4#	5#	6#	7#	8#	9#
V Ni Cr Co	$\begin{array}{c} 0.017 \pm 0.002 \\ 0.004 \pm 0.001 \\ 0.018 \pm 0.002 \\ 0.006 \pm 0.001 \end{array}$	$\begin{array}{c} 0.015 \pm 0.001 \\ 0.004 \pm 0.002 \\ 0.013 \pm 0.001 \\ 0.005 \pm 0.001 \end{array}$	$\begin{array}{c} 0.016 \pm 0.001 \\ 0.005 \pm 0.001 \\ 0.014 \pm 0.001 \\ 0.002 \pm 0.001 \end{array}$	$\begin{array}{c} 0.019 \pm 0.002 \\ 0.001 \pm 0.001 \\ 0.009 \pm 0.001 \\ 0.004 \pm 0.001 \end{array}$	$\begin{array}{c} 0.015 \pm 0.002 \\ - \\ 0.016 \pm 0.002 \\ 0.002 \pm 0.001 \end{array}$	$\begin{array}{c} 0.023 \pm 0.002 \\ 0.009 \pm 0.002 \\ 0.017 \pm 0.002 \\ 0.005 \pm 0.001 \end{array}$	$\begin{array}{c} 0.025 \pm 0.001 \\ 0.013 \pm 0.002 \\ 0.021 \pm 0.001 \\ 0.007 \pm 0.001 \end{array}$	$\begin{array}{c} 0.022 \pm 0.001 \\ 0.006 \pm 0.001 \\ 0.021 \pm 0.001 \\ 0.007 \pm 0.001 \end{array}$	$\begin{array}{c} 0.020 \pm 0.001 \\ 0.013 \pm 0.001 \\ 0.021 \pm 0.001 \\ 0.006 \pm 0.001 \end{array}$
Sample No.	10#	11#	12#	13#	14#	15#	16#	17#	18#
Si Ca Fe Al K Mg Ti Ba Na Zr Zn Zn P Mn	$\begin{array}{c} 18.93 \pm 0.11\\ 5.47 \pm 0.09\\ 10.94 \pm 0.13\\ 7.49 \pm 0.09\\ 5.93 \pm 0.11\\ 1.67 \pm 0.05\\ 0.71 \pm 0.03\\ 0.27 \pm 0.01\\ 0.22 \pm 0.02\\ 0.15 \pm 0.01\\ 0.02 \pm <0.01\\ 0.04 \pm <0.01\\ 0.059 \pm 0.003\\ 0.023 \pm 0.003\\ \end{array}$	$\begin{array}{c} 20.12\pm0.12\\ 4.01\pm0.08\\ 10.21\pm0.12\\ 5.97\pm0.11\\ 5.97\pm0.11\\ 1.72\pm0.05\\ 0.74\pm0.03\\ 0.29\pm0.01\\ 0.23\pm0.02\\ 0.17\pm0.01\\ 0.23\pm0.01\\ 0.04\pm<0.01\\ 0.054\pm0.003\\ 0.054\pm0.003\\ 0.054\pm0.003\\ \end{array}$	$\begin{array}{c} 19.04\pm 0.11\\ 8.52\pm 0.12\\ 8.16\pm 0.11\\ 7.03\pm 0.09\\ 5.01\pm 0.10\\ 1.58\pm 0.05\\ 0.76\pm 0.03\\ 0.20\pm 0.01\\ 0.35\pm 0.02\\ 0.27\pm 0.01\\ 0.23\pm 0.01\\ 0.04\pm <0.01\\ 0.060\pm 0.003\\ 0.0$	$\begin{array}{c} 16.50 \pm 0.11 \\ 9.52 \pm 0.12 \\ 9.30 \pm 0.12 \\ 7.39 \pm 0.09 \\ 5.48 \pm 0.10 \\ 1.76 \pm 0.05 \\ 0.88 \pm 0.04 \\ 0.34 \pm 0.02 \\ 0.37 \pm 0.02 \\ 0.21 \pm 0.01 \\ 0.17 \pm 0.01 \\ 0.04 \pm < 0.01 \\ 0.076 \pm 0.004 \\ 0.002 \\ 0.024 \pm 0.001 \\ 0.076 \pm 0.004 \\ 0.002 \\ 0.002 \\ 0.002 \\ 0.002 \\ 0.002 \\ 0.002 \\ 0.002 \\ 0.0004 \\ 0$	$\begin{array}{c} 18.18 \pm 0.11 \\ 7.91 \pm 0.11 \\ 9.90 \pm 0.12 \\ 7.12 \pm 0.09 \\ 5.11 \pm 0.10 \\ 1.73 \pm 0.05 \\ 0.79 \pm 0.03 \\ 0.18 \pm 0.01 \\ 0.47 \pm 0.02 \\ 0.30 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.05 \pm 0.01 \\ 0.051 \pm 0.003 \\ 0.021 \\ 0.001 \\ 0.001 \\ 0.001 \\ 0.001 \\ 0.001 \\ 0.0001 \\ 0.0001 \\ 0.0001 \\ 0.0001 \\ 0.0001 \\ 0.0001 \\ 0.0001 \\ 0.0001 \\ 0.0001 \\ 0.0001 \\ 0.00000 \\ 0.0001 \\ 0.00000 \\ 0.0000 $	$\begin{array}{c} 24.54 \pm 0.12 \\ 0.40 \pm 0.02 \\ 9.90 \pm 0.12 \\ 6.40 \pm 0.09 \\ 5.02 \pm 0.10 \\ 1.15 \pm 0.04 \\ 0.60 \pm 0.03 \\ 0.20 \pm 0.01 \\ 0.11 \pm 0.01 \\ 0.12 \pm 0.01 \\ 0.06 \pm < 0.01 \\ 0.07 \pm < 0.01 \\ 0.025 \pm 0.001 \\ 0.025 \pm 0.001 \end{array}$	$\begin{array}{c} 23.82 \pm 0.12 \\ 0.83 \pm 0.04 \\ 10.57 \pm 0.13 \\ 6.15 \pm 0.08 \\ 4.94 \pm 0.10 \\ 1.26 \pm 0.04 \\ 0.58 \pm 0.03 \\ 0.26 \pm 0.01 \\ 0.09 \pm 0.01 \\ 0.15 \pm 0.01 \\ 0.06 \pm < 0.01 \\ 0.06 \pm < 0.01 \\ 0.030 \pm 0.002 \\ 0.030 \pm 0.002 \end{array}$	$\begin{array}{c} 29.14\pm 0.11\\ 0.47\pm 0.02\\ 6.59\pm 0.10\\ 5.12\pm 0.08\\ 4.24\pm 0.09\\ 0.91\pm 0.04\\ 0.49\pm 0.02\\ 0.09\pm <0.01\\ 0.16\pm 0.01\\ 0.10\pm <0.01\\ 0.21\pm 0.01\\ 0.06\pm <0.01\\ 0.027\pm 0.002\\ 0.027\pm 0.002\\ 0.027\pm 0.002\\ \end{array}$	$\begin{array}{c} 29.92\pm0.11\\ 0.23\pm0.01\\ 6.60\pm0.10\\ 4.81\pm0.08\\ 3.87\pm0.09\\ 0.84\pm0.04\\ 0.47\pm0.02\\ 0.16\pm0.01\\ 0.13\pm0.01\\ 0.12\pm0.01\\ 0.03\pm<0.01\\ 0.07\pm<0.01\\ 0.021\pm0.001\\ 0.021\pm0.001\end{array}$
v Ni	0.023 ± 0.001 0.014 ± 0.002	0.024 ± 0.001 0.010 ± 0.001	0.023 ± 0.002 0.010 ± 0.002	0.022 ± 0.002 0.006 ± 0.002	$\begin{array}{c} 0.021 \pm 0.002 \\ 0.008 \pm 0.002 \end{array}$	0.039 ± 0.002 0.043 ± 0.002	$\begin{array}{c} 0.036 \pm 0.002 \\ 0.049 \pm 0.003 \end{array}$	$\begin{array}{c} 0.034 \pm 0.002 \\ 0.019 \pm 0.002 \end{array}$	$\begin{array}{c} 0.032 \pm 0.002 \\ 0.022 \pm 0.001 \end{array}$
Cr Co	$\begin{array}{c} 0.020 \pm 0.001 \\ 0.005 \pm 0.001 \end{array}$	$\begin{array}{c} 0.020 \pm 0.001 \\ 0.005 \pm 0.001 \end{array}$	$\begin{array}{c} 0.015 \pm 0.001 \\ 0.006 \pm 0.001 \end{array}$	$\begin{array}{c} 0.023 \pm 0.002 \\ 0.004 \pm 0.002 \end{array}$	$\begin{array}{c} 0.016 \pm 0.002 \\ 0.006 \pm 0.002 \end{array}$	$\begin{array}{c} 0.015 \pm 0.001 \\ 0.005 \pm 0.002 \end{array}$	$\begin{array}{c} 0.015 \pm 0.001 \\ 0.008 \pm 0.001 \end{array}$	$\begin{array}{c} 0.013 \pm 0.001 \\ 0.001 \pm 0.001 \end{array}$	$\begin{array}{c} 0.013 \pm 0.001 \\ 0.005 \pm 0.001 \end{array}$

Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.geoen.2024.212777.

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