

It's cute (可爱).....

But does  
it DO  
anything??



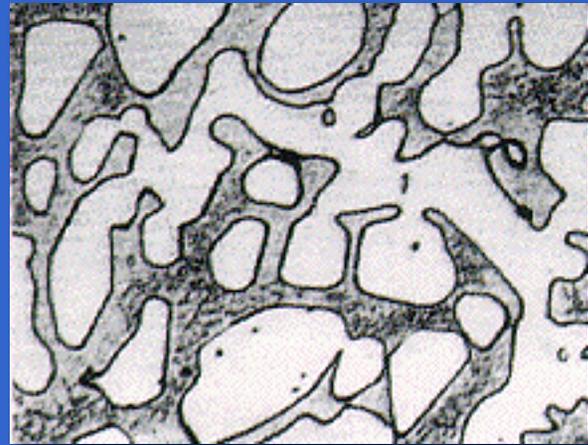
# Powder diffraction - applications

Qualitative analysis - what IS this stuff?!!

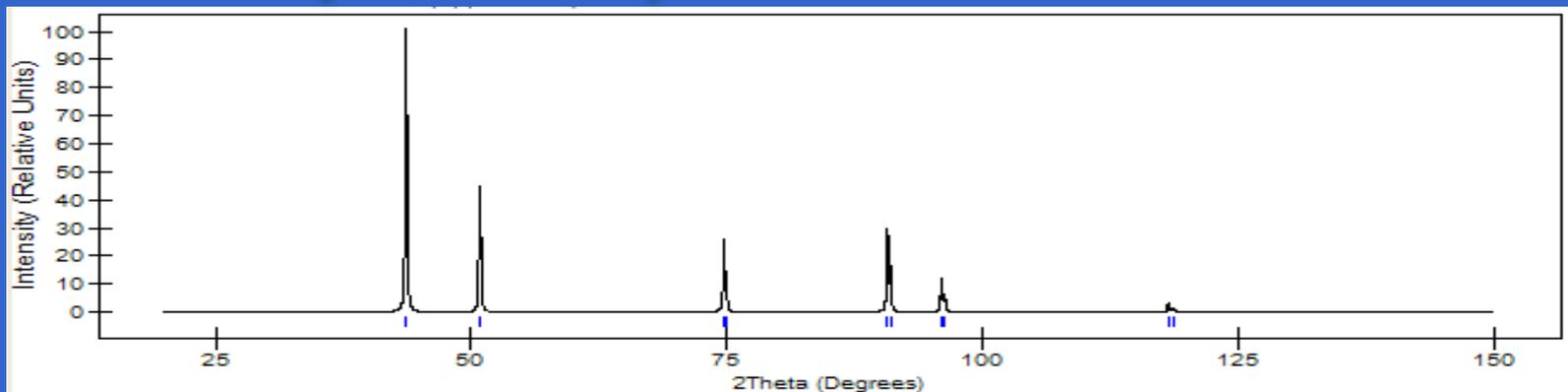
Every material has a unique x-ray powder pattern  
Every PHASE has a unique powder pattern

This image, taken under a microscope, shows that a 'duplex stainless steel' has two phases

双相不锈钢

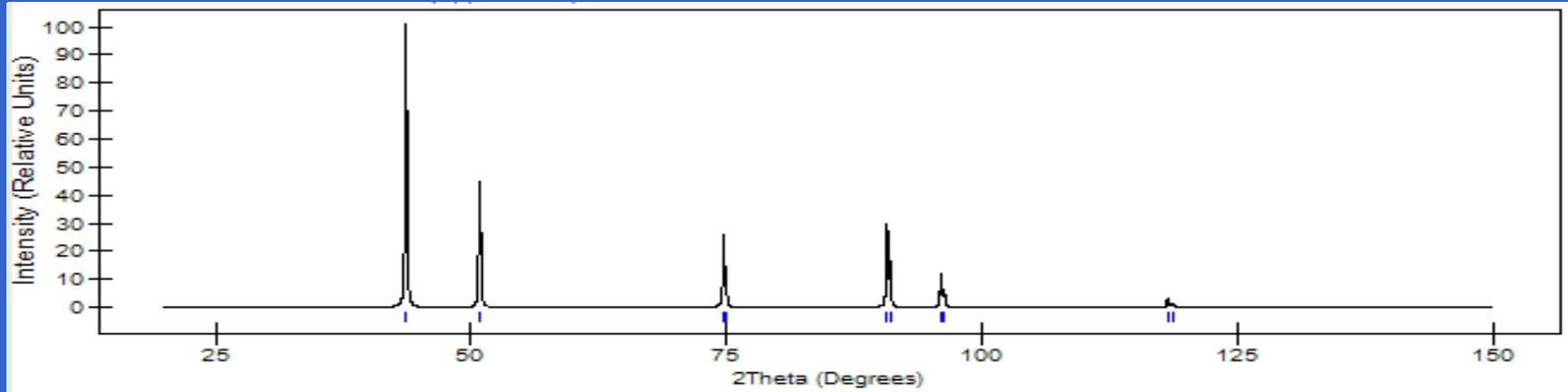


This is the X-ray diffraction pattern



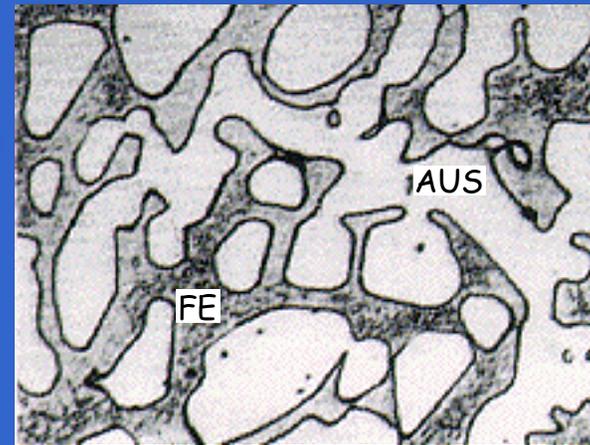
# Powder diffraction - applications

Qualitative analysis - what IS this stuff?!!



The X-ray diffraction pattern tells us this

How?



# Powder diffraction - applications

Qualitative analysis - what IS this stuff?!!

We use the Powder Diffraction File database (the PDF)  
and a search system

## History

1938 Hanawalt paper on first database and search method

1941 ASTM committee publishes database on cards

1969 JCPDS incorporated; publishes the JCPDF

# Powder diffraction - applications

Qualitative analysis - what IS this stuff?!!

We use the Powder Diffraction File database (the PDF)  
and a search system

## History

1938 Hanawalt paper on first database and search method

1941 ASTM committee publishes database on cards

1969 JCPDS incorporated; publishes the JCPDF

1977 JCPDS became the International Centre for  
Diffraction Data (ICDD) - currently publishing various  
databases, including the new PDF-4, with diffraction  
and structure data for >300,000 phases



d-Spacings

Wavelength

Cu Kα1 1.54056Å

Å

Slit Pattern

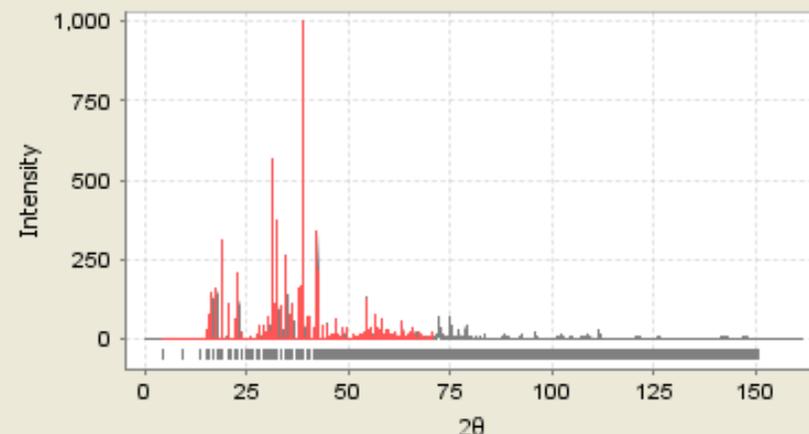
- Fixed Slit Intensity
- Variable Slit Intensity
- Integrated Intensity

Diffraction Pattern

- Calculated Pattern
- PD3 Pattern

Fixed Slit Intensity

2θ	d(Å)	I	h	k	l	*
4.5511	19.400000	1	2	0	0	
9.1093	9.700000	1	4	0	0	
13.6821	6.466670	1	6	0	0	
15.1921	5.827150	32	1	0	1	
15.7008	5.639470	75	2	0	1	
16.5147	5.363320	145	3	0	1	
17.5928	5.037030	163	4	0	1	
18.8913	4.693630	314	5	0	1	
20.3701	4.356100	113	6	0	1	
21.9950	4.037830	61	7	0	1	
22.9015	3.880000	206	10	0	0	



— 04-007-1562 (Fixed Slit Intensity) — 04-007-1562 (Calculated)

PDF #: 04-007-1562

Status: Primary

QM: Star (S)

Pressure/Temperature: Ambient

Chemical Formula: Ca Cu

Structural Formula:

Empirical Formula: Ca Cu

Weight %: Ca38.68 Cu61.32

Atomic %: Ca50.00 Cu50.00

ANX: NO

Compound Name: Calcium Copper

Mineral Name:

Common Name: α-Ca Cu



d-Spacings

Wavelength

Cu Kα1 1.54056Å

Å

Stick Pattern

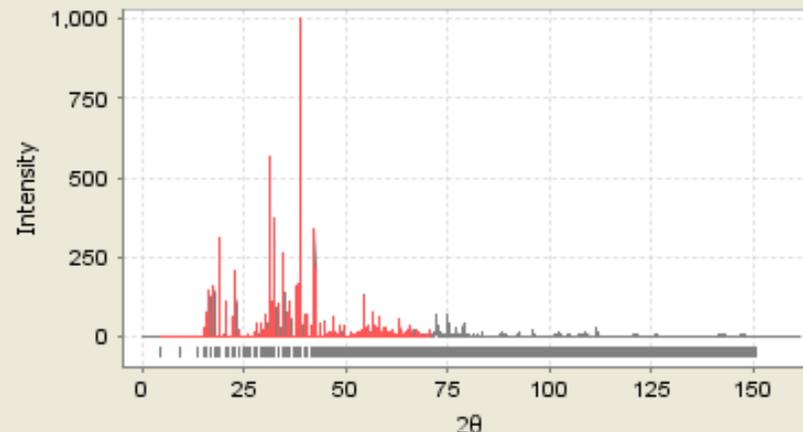
- Fixed Slit Intensity
- Variable Slit Intensity
- Integrated Intensity

Diffraction Pattern

- Calculated Pattern
- PD3 Pattern

Fixed Slit Intensity

2θ	d(Å)	I	h	k	l	*
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13.6821	6.466670	1	6	0	0	
15.1921	5.827150	32	1	0	1	
15.7008	5.639470	75	2	0	1	
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17.5928	5.037030	163	4	0	1	
18.8913	4.693630	314	5	0	1	
20.3701	4.356100	113	6	0	1	
21.9950	4.037830	61	7	0	1	
22.9015	3.880000	206	10	0	0	



— 04-007-1562 (Fixed Slit Intensity) — 04-007-1562 (Calculated)

Author's Reported Data (See Reference tab for source)

Radiation: CuKα1

λ: 1.5406 Å

Filter:

d-Spacing: Calculated

Cutoff:

Intensity: Calculated

I/Ic: 2.71

Camera Diameter:



d-Spacings

Wavelength

Cu Kα1 1.54056Å

Å

Stick Pattern

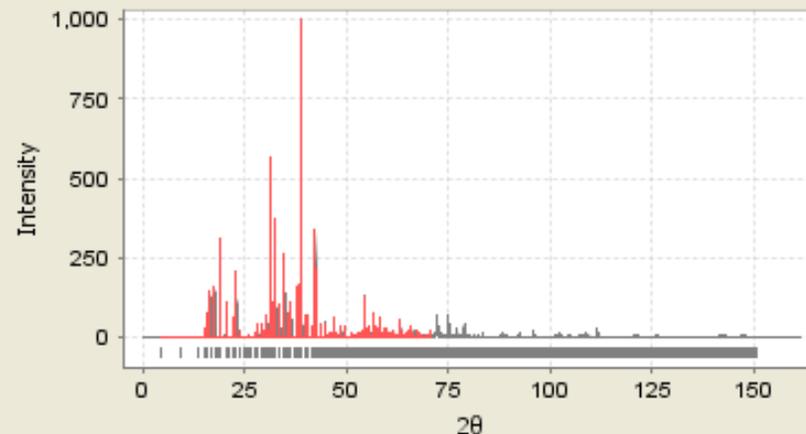
- Fixed Slit Intensity
- Variable Slit Intensity
- Integrated Intensity

Diffraction Pattern

- Calculated Pattern
- PD3 Pattern

Fixed Slit Intensity

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15.1921	5.827150	32	1	0	1	
15.7008	5.639470	75	2	0	1	
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18.8913	4.693630	314	5	0	1	
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21.9950	4.037830	61	7	0	1	
22.9015	3.880000	206	10	0	0	



— 04-007-1562 (Fixed Slit Intensity) — 04-007-1562 (Calculated)

SYS: Orthorhombic

Space Group: Pnma (62)

Aspect:

Author's Cell (See Reference tab for source)

a: 38.80(1) Å

b: 4.2710(20) Å

c: 5.8940(90) Å

α: °

β: °

γ: °

Volume: 976.72 Å<sup>3</sup>

Z: 20.00

MolVol: 48.84



Author's Cell Axial Ratio

c/a: 0.152

a/b: 9.085

c/b: 1.380

Dcalc: 3.524 g/cm<sup>3</sup>

Dmeas: g/cm<sup>3</sup>

Dstruc: 3.52 g/cm<sup>3</sup>

SS/FOM: F(30) = 148.2(0.0051, 40)

Melting Point:

R-factor: 0.061

Error:



d-Spacings

Wavelength

Cu Kα1 1.54056Å

Å

Slit Pattern

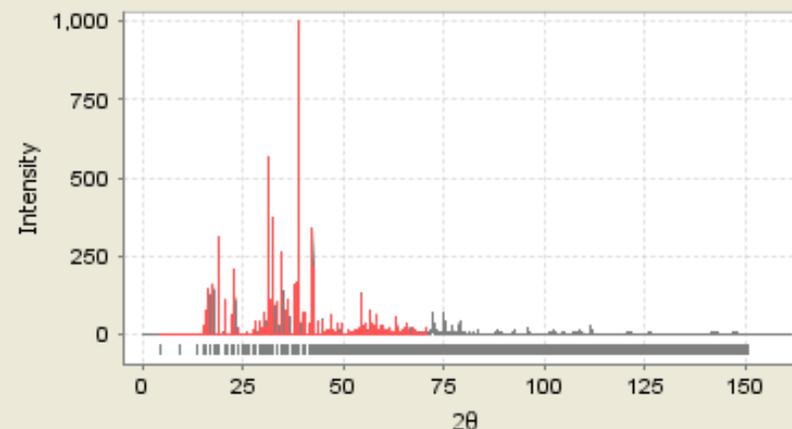
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- Variable Slit Intensity
- Integrated Intensity

Diffraction Pattern

- Calculated Pattern
- PD3 Pattern

Fixed Slit Intensity

2θ	d(Å)	I	h	k	l	*
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18.8913	4.693630	314	5	0	1	
20.3701	4.356100	113	6	0	1	
21.9950	4.037830	61	7	0	1	
22.9015	3.880000	206	10	0	0	



— 04-007-1562 (Fixed Slit Intensity) — 04-007-1562 (Calculated)

ICDD Calculated Parameters

Space Group: Pbnm (62)

Molecular Weight: 103.63

Crystal Data

a: 5.894 Å      b: 38.800 Å      c: 4.271 Å  
 α: 90.00 °      β: 90.00 °      γ: 90.00 °  
 Volume: 976.72 Å<sup>3</sup>      z: 20.00

Crystal Data Axial Ratio

c/a: 0.725      a/b: 0.152      c/b: 0.110

Reduced Cell

a: 4.271 Å      b: 5.894 Å      c: 38.800 Å  
 α: 90.00 °      β: 90.00 °      γ: 90.00 °  
 Volume: 976.72 Å<sup>3</sup>



d-Spacings

Wavelength

Cu Ka1 1.54056Å

Å

Stick Pattern

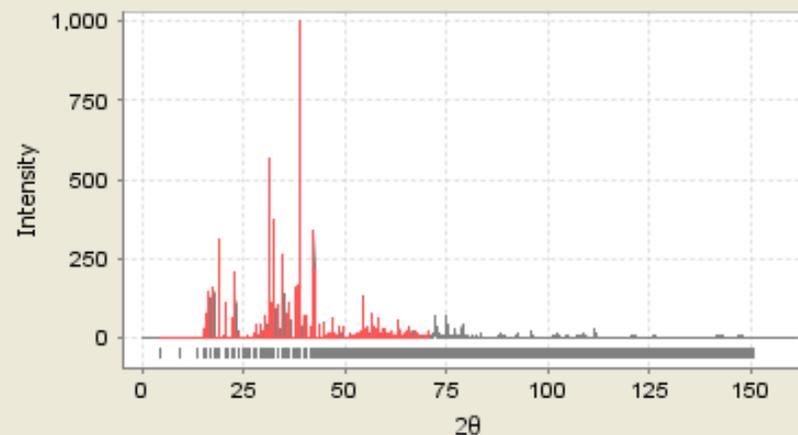
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- Variable Slit Intensity
- Integrated Intensity

Diffraction Pattern

- Calculated Pattern
- PD3 Pattern

Fixed Slit Intensity

2θ	d(Å)	I	h	k	l	*
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15.1921	5.827150	32	1	0	1	
15.7008	5.639470	75	2	0	1	
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18.8913	4.693630	314	5	0	1	
20.3701	4.356100	113	6	0	1	
21.9950	4.037830	61	7	0	1	
22.9015	3.880000	206	10	0	0	



— 04-007-1562 (Fixed Slit Intensity) — 04-007-1562 (Calculated)

(See Reference tab for source)

ca:

nωβ:

εγ:

Sign:

2V:



d-Spacings

Wavelength

Cu Kα1 1.54056Å

Å

Stick Pattern

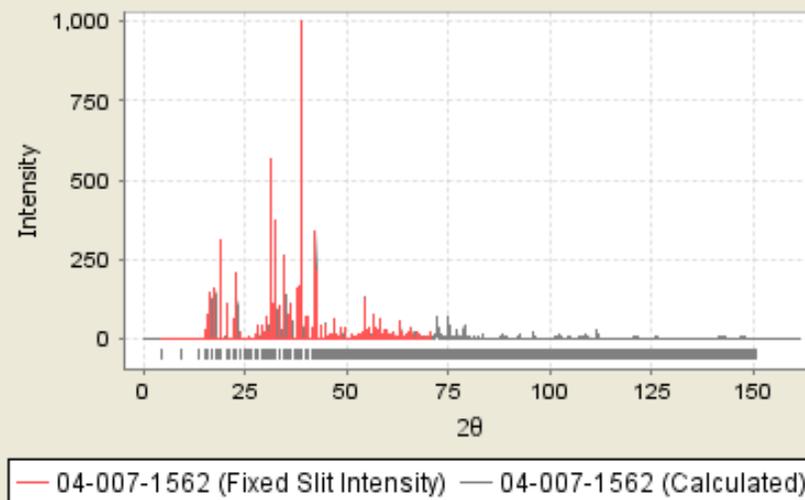
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Diffraction Pattern

- Calculated Pattern
- PD3 Pattern

Fixed Slit Intensity

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21.9950	4.037830	61	7	0	1	
22.9015	3.880000	206	10	0	0	



SG Symmetry Operators (8)

Seq	Operator
1	x,y,z
2	-x,-y,-z
3	x+1/2,-y+1/2,-z+1/2
4	-x+1/2,y+1/2,z+1/2
5	-x,y+1/2,-z
6	x,-y+1/2,z
7	-x+1/2,-y,z+1/2
8	x+1/2,y,-z+1/2

TDP Type: U Origin: Crystal (Symmetry Allowed): Centrosymmetric

Atomic Coordinates (10)

Atom	Num	Wyckoff	Symmetry	x	y	z	SOF	IDP	AET
Ca	1	4c	.m.	0.4638	0.25	0.6321	1.0	0.02	17-d
Ca	2	4c	.m.	0.3652	0.25	0.8735	1.0	0.02	17-d
Ca	3	4c	.m.	0.2639	0.25	0.1259	1.0	0.02	17-d
Ca	4	4c	.m.	0.1644	0.25	0.8726	1.0	0.02	17-d

Anisotropic Thermal Displacement Parameters (0)

Anisotropic Thermal Displacement Parameters (0)									

PDF Card - Ca Cu - 04-007-1562

File Edit d-Spacings Tools Window Help



d-Spacings

Wavelength

Cu Kα1 1.54056Å

Å

Stick Pattern

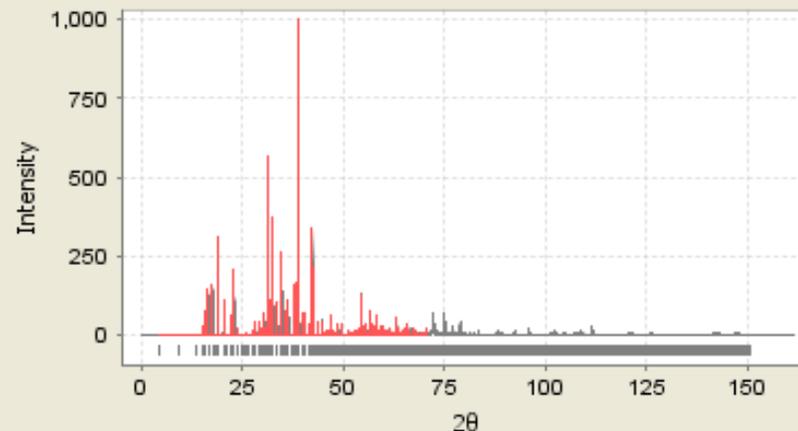
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- Variable Slit Intensity
- Integrated Intensity

Diffraction Pattern

- Calculated Pattern
- PD3 Pattern

Fixed Slit Intensity

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15.7008	5.639470	75	2	0	1	
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21.9950	4.037830	61	7	0	1	
22.9015	3.880000	206	10	0	0	



— 04-007-1562 (Fixed Slit Intensity) — 04-007-1562 (Calculated)

PDF Experimental Physical Crystal Optical Structure Miscellaneous References Comments

CAS:

Former PDF's #: 01-072-5678, 03-065-2256, 03-065-7630

Pearson: oP40.00

Cross-Ref PDF #'s (1)

00-041-1273 (Primary)

Pearson w/o H:

Prototype Structure:

Prototype Structure (Alpha Order):

LPF Prototype Structure: Ca Cu<sub>2</sub>oP40,62

LPF Prototype Structure (Alpha Order): Ca Cu

Mineral Classification:

Zeolite Classification:

Subfile(s): LPF Pattern, Metals & Alloys, Primary Pattern, Inorganic

Entry Date: 01/05/2005

Last Modification Date: 01/25/2010



d-Spacings

Wavelength

Cu Kα1 1.54056Å

Å

Slit Pattern

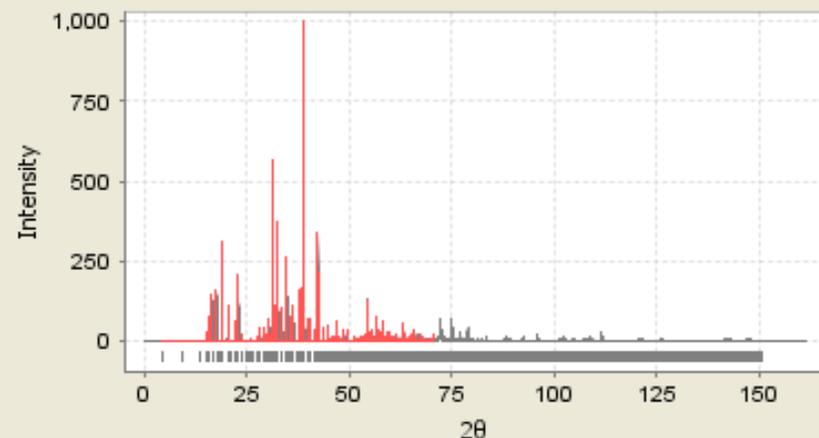
- Fixed Slit Intensity
- Variable Slit Intensity
- Integrated Intensity

Diffraction Pattern

- Calculated Pattern
- PD3 Pattern

Fixed Slit Intensity

2θ	d(Å)	I	h	k	l	*
4.5511	19.400000	1	2	0	0	
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15.7008	5.639470	75	2	0	1	
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18.8913	4.693630	314	5	0	1	
20.3701	4.356100	113	6	0	1	
21.9950	4.037830	61	7	0	1	
22.9015	3.880000	206	10	0	0	



— 04-007-1562 (Fixed Slit Intensity) — 04-007-1562 (Calculated)

Experimental Data Reference

"The Structures of  $\alpha$ -CaCu,  $\beta$ -CaCu, SrAg and BaAg: Four Different Stacking Variants Based on Noble-Metal-Centred Trigonal Prisms". Merlo F., Fornasini M.L. *Acta Crystallogr., Sec. B: Struct. Crystallogr. Cryst. Chem.* **37**, 500 (1981). Calculated from LPF using POWD-12++.

Additional Powder Data Reference(s)

Physical Data Reference(s)

Ibid.

Optical Data Reference(s)



d-Spacings

Wavelength

Cu Kα1 1.54056Å

Å

Slit Pattern

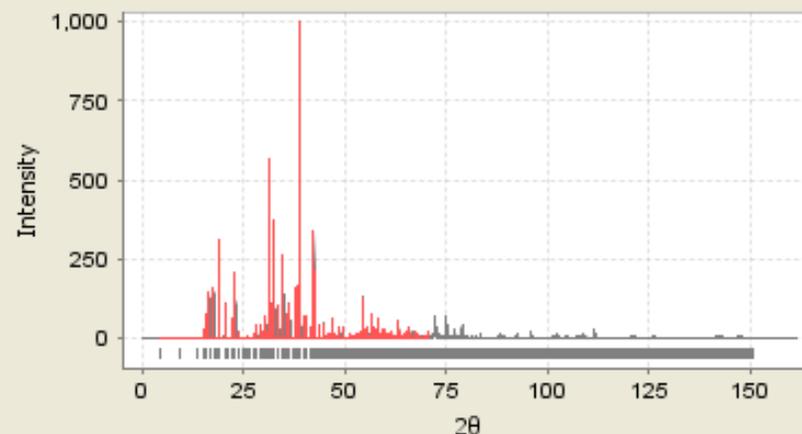
- Fixed Slit Intensity
- Variable Slit Intensity
- Integrated Intensity

Diffraction Pattern

- Calculated Pattern
- PD3 Pattern

Fixed Slit Intensity

2θ	d(Å)	I	h	k	l	*
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9.1093	9.700000	1	4	0	0	
13.6821	6.466670	1	6	0	0	
15.1921	5.827150	32	1	0	1	
15.7008	5.639470	75	2	0	1	
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17.5928	5.037030	163	4	0	1	
18.8913	4.693630	314	5	0	1	
20.3701	4.356100	113	6	0	1	
21.9950	4.037830	61	7	0	1	
22.9015	3.880000	206	10	0	0	



Database Comments

ANX: NO. LPF Collection Code: 1251296. Sample Preparation: STARTING MATERIALS:Ca,Cu. Compound Preparation: melted, water-quenched from 873 K. CRUCIBLE: iron. ATMOSPHERE: argon. Unit Cell Data Source: Single Crystal.

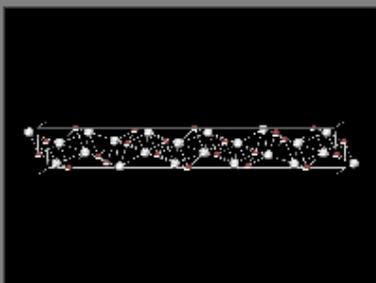
User Comments

Empty text input field for user comments.

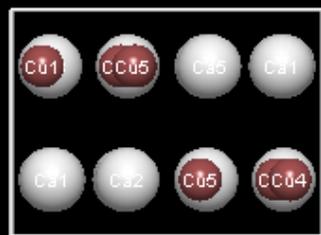
Save Comment

Shared Comments

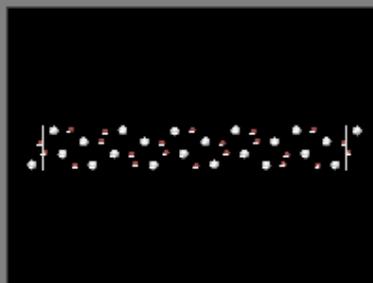
Empty text input field for shared comments.



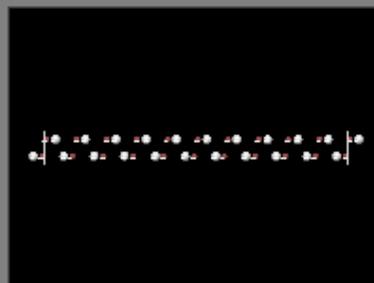
3D cell projection



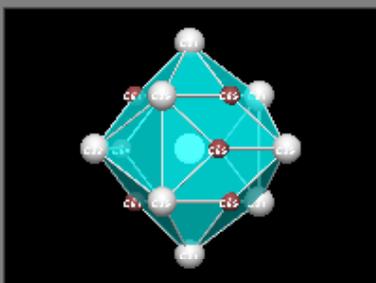
2D cell projection (100)



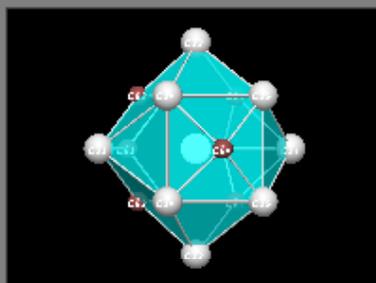
2D cell projection (010)



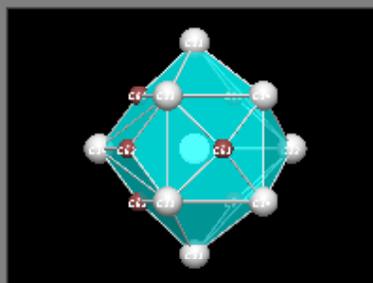
2D cell projection (001)



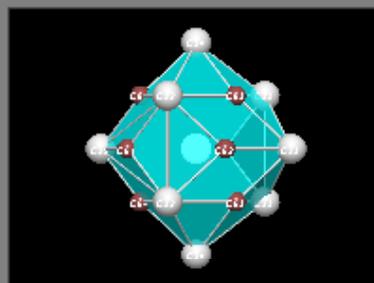
Atomic environment of Ca1



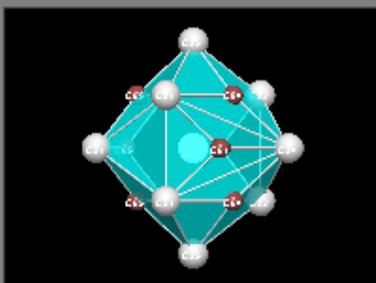
Atomic environment of Ca2



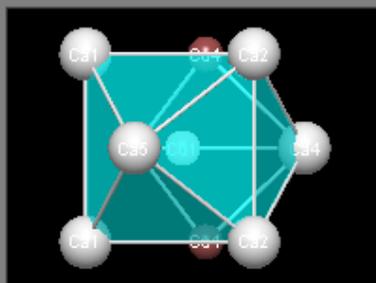
Atomic environment of Ca3



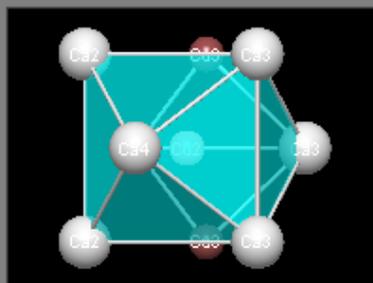
Atomic environment of Ca4



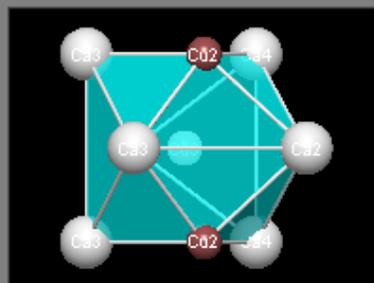
Atomic environment of Ca5



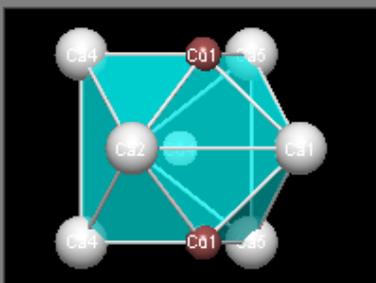
Atomic environment of Cu1



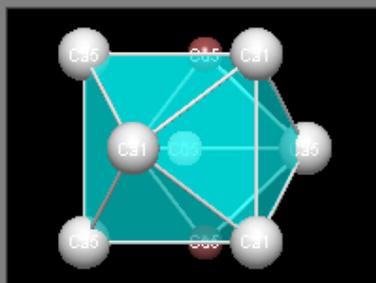
Atomic environment of Cu2



Atomic environment of Cu3



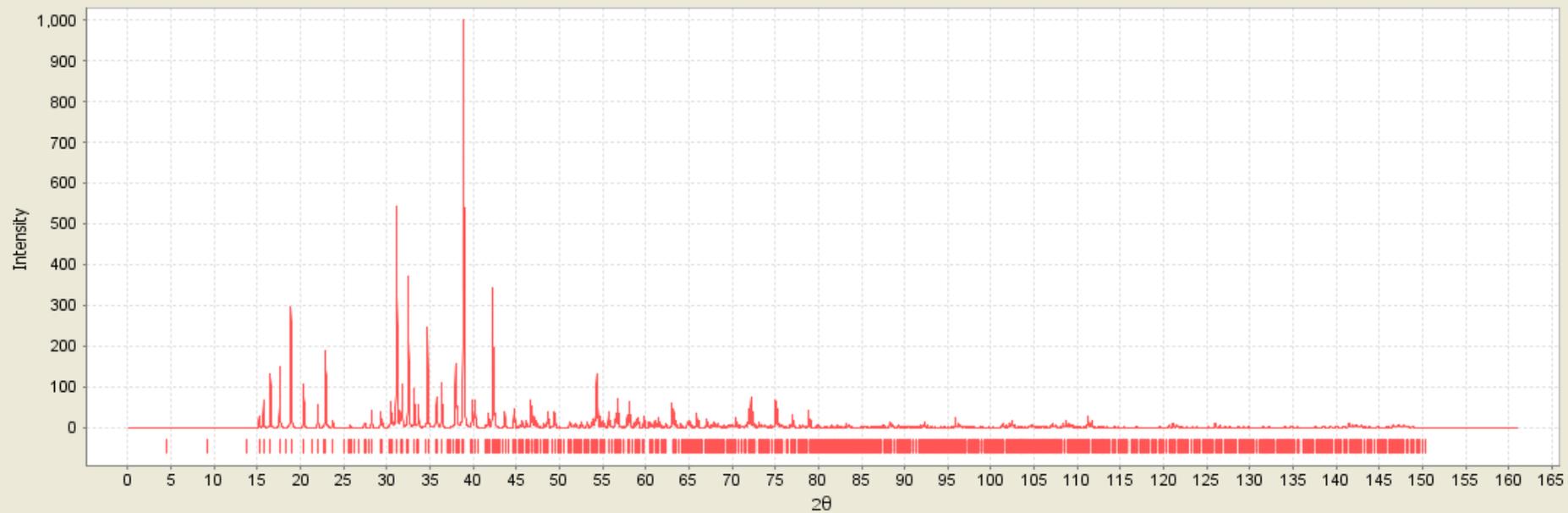
Atomic environment of Cu4



Atomic environment of Cu5

Diffraction Pattern - Ca Cu - 04-007-1562 (Calculated)

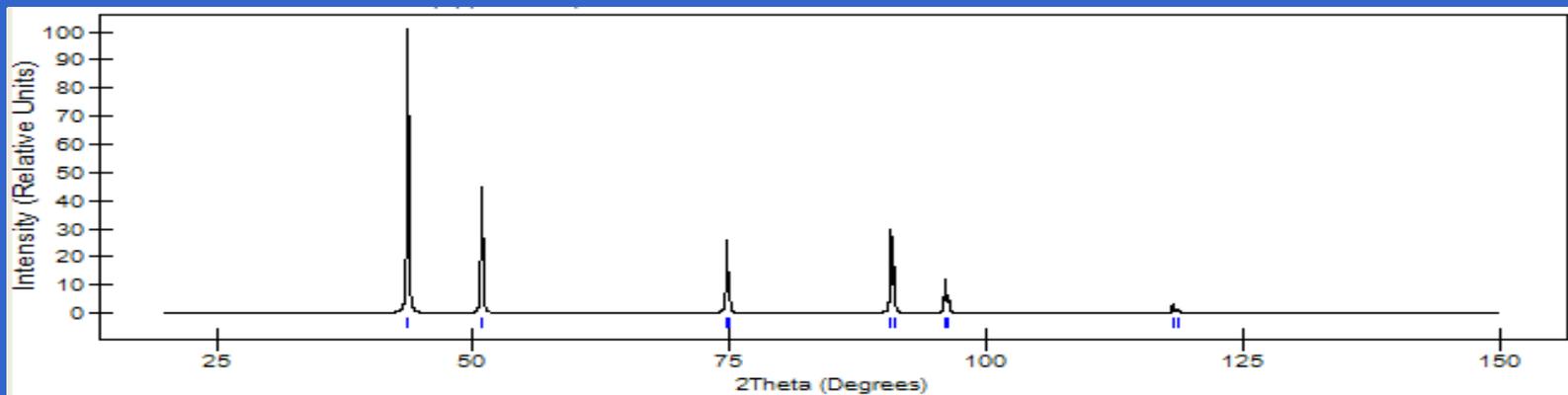
File Edit Plots Window Help



Ca Cu - 04-007-1562 (Calculated)

# Powder diffraction - applications

Qualitative analysis - what IS this stuff?!!



Make d, I list:

2.080	100	1
2.025	100	2
1.798	45	4
1.433	20	
1.272	26	7
1.170	50	3

Do search/match

1.081	30	6
1.040	12	
1.018	20	8
0.907	35	5
0.898	3	
0.830	14	

# Powder diffraction - applications

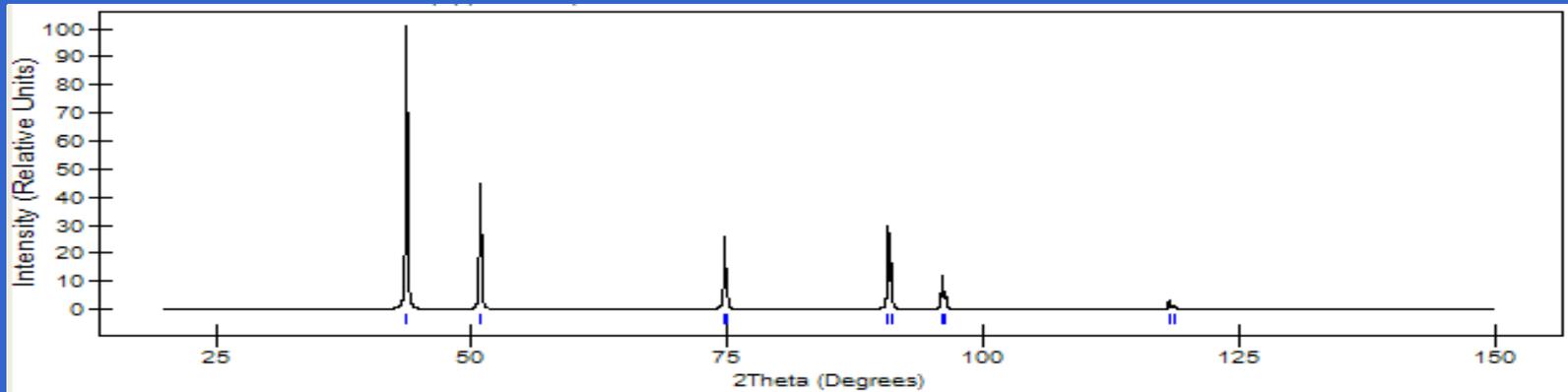
Qualitative analysis - what IS this stuff?!!

Example of page of search manual for Hanawalt manual search/match

3.39 - 3.32 ( $\pm .02$ )								File No.	
3.38 <sub>y</sub>	8.58 <sub>x</sub>	3.04 <sub>y</sub>	4.11 <sub>z</sub>	3.18 <sub>z</sub>	1.69 <sub>z</sub>	2.65 <sub>z</sub>	1.88 <sub>z</sub>	(Mg,Fe) <sub>2</sub> Al <sub>2</sub> Si <sub>2</sub> O <sub>10</sub> /Cordierite, ferroan	9- 472
3.33 <sub>x</sub>	6.72 <sub>y</sub>	3.19 <sub>z</sub>	8.09 <sub>z</sub>	3.28 <sub>z</sub>	5.18 <sub>z</sub>	3.10 <sub>z</sub>	4.30 <sub>z</sub>	C <sub>10</sub> H <sub>10</sub> N <sub>2</sub> O <sub>6</sub>	29-1716
3.31 <sub>z</sub>	6.40 <sub>x</sub>	6.10 <sub>z</sub>	3.85 <sub>z</sub>	2.77 <sub>z</sub>	6.70 <sub>z</sub>	3.48 <sub>z</sub>	2.64 <sub>z</sub>	C <sub>12</sub> H <sub>6</sub> Cl <sub>6</sub>	17-1054
3.38 <sub>y</sub>	6.13 <sub>x</sub>	8.66 <sub>y</sub>	3.20 <sub>y</sub>	3.29 <sub>z</sub>	9.70 <sub>z</sub>	4.57 <sub>z</sub>	3.46 <sub>z</sub>	C <sub>11</sub> H <sub>11</sub> N <sub>5</sub> -HCl	28-1749
3.34 <sub>x</sub>	5.93 <sub>z</sub>	5.19 <sub>z</sub>	3.77 <sub>z</sub>	3.65 <sub>z</sub>	3.51 <sub>z</sub>	2.94 <sub>z</sub>	1.67 <sub>z</sub>	C <sub>4</sub> H <sub>6</sub> N <sub>2</sub> O <sub>2</sub>	26-1863
3.37 <sub>x</sub>	5.85 <sub>z</sub>	3.86 <sub>z</sub>	3.72 <sub>z</sub>	3.52 <sub>z</sub>	3.03 <sub>z</sub>	2.70 <sub>z</sub>	7.72 <sub>z</sub>	C <sub>6</sub> H <sub>6</sub> N <sub>2</sub> O <sub>2</sub> -HCl	5- 459
3.31 <sub>z</sub>	5.73 <sub>x</sub>	3.43 <sub>z</sub>	3.59 <sub>z</sub>	3.19 <sub>z</sub>	4.36 <sub>z</sub>	4.19 <sub>z</sub>	3.27 <sub>z</sub>	C <sub>6</sub> H <sub>5</sub> NO <sub>2</sub>	30-1845
3.30 <sub>x</sub>	5.44 <sub>z</sub>	5.63 <sub>z</sub>	3.24 <sub>z</sub>	4.97 <sub>z</sub>	6.58 <sub>z</sub>	4.58 <sub>z</sub>	3.15 <sub>z</sub>	(NH <sub>4</sub> ) <sub>2</sub> P <sub>2</sub> O <sub>7</sub>	20- 102
3.38 <sub>x</sub>	5.30 <sub>x</sub>	3.49 <sub>x</sub>	5.90 <sub>z</sub>	3.67 <sub>z</sub>	3.26 <sub>z</sub>	3.18 <sub>z</sub>	2.99 <sub>z</sub>	KH <sub>2</sub> P <sub>2</sub> O <sub>7</sub>	15- 509
3.35 <sub>x</sub>	5.21 <sub>z</sub>	4.86 <sub>z</sub>	4.33 <sub>z</sub>	4.04 <sub>z</sub>	3.90 <sub>z</sub>	3.55 <sub>z</sub>	2.73 <sub>z</sub>	$\beta$ -C <sub>7</sub> H <sub>11</sub> NO <sub>2</sub>	22-1874
3.40 <sub>x</sub>	5.01 <sub>y</sub>	3.09 <sub>z</sub>	4.10 <sub>z</sub>	3.00 <sub>z</sub>	4.03 <sub>z</sub>	6.74 <sub>z</sub>	3.45 <sub>z</sub>	C <sub>7</sub> H <sub>6</sub> N <sub>6</sub>	24-1654
3.30 <sub>x</sub>	4.76 <sub>z</sub>	4.18 <sub>z</sub>	5.73 <sub>z</sub>	2.92 <sub>z</sub>	3.98 <sub>z</sub>	2.38 <sub>z</sub>	3.35 <sub>z</sub>	C <sub>6</sub> H <sub>4</sub> O <sub>4</sub>	37-1919
3.31 <sub>x</sub>	4.71 <sub>z</sub>	3.50 <sub>z</sub>	5.56 <sub>z</sub>	3.84 <sub>z</sub>	3.03 <sub>z</sub>	7.02 <sub>z</sub>	2.30 <sub>z</sub>	C <sub>6</sub> H <sub>5</sub> NO <sub>2</sub> -HCl	29-1827
3.39 <sub>x</sub>	4.48 <sub>z</sub>	3.43 <sub>z</sub>	3.01 <sub>z</sub>	4.09 <sub>z</sub>	2.98 <sub>z</sub>	2.78 <sub>z</sub>	3.18 <sub>z</sub>	NaHSO <sub>4</sub>	25- 833
3.34 <sub>y</sub>	4.42 <sub>x</sub>	10.1 <sub>y</sub>	1.48 <sub>y</sub>	2.56 <sub>z</sub>	1.68 <sub>z</sub>	1.28 <sub>z</sub>	1.23 <sub>z</sub>	Al <sub>2</sub> Si <sub>2</sub> O <sub>7</sub> (OH) <sub>4</sub> ·2H <sub>2</sub> O/Holloyite-10A	9- 451
3.40 <sub>y</sub>	4.38 <sub>x</sub>	2.88 <sub>z</sub>	5.76 <sub>z</sub>	2.61 <sub>z</sub>	4.09 <sub>z</sub>	2.76 <sub>z</sub>	1.76 <sub>z</sub>	V <sub>2</sub> O <sub>5</sub> /Shcherbinoite, syn	9- 387
3.33 <sub>x</sub>	4.30 <sub>z</sub>	2.82 <sub>z</sub>	6.08 <sub>z</sub>	4.72 <sub>z</sub>	1.71 <sub>z</sub>	3.52 <sub>z</sub>	2.15 <sub>z</sub>	(NH <sub>4</sub> ) <sub>2</sub> Co <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	22-1037
3.37 <sub>x</sub>	4.28 <sub>z</sub>	1.84 <sub>z</sub>	1.55 <sub>z</sub>	2.47 <sub>z</sub>	2.31 <sub>z</sub>	1.39 <sub>z</sub>	2.14 <sub>z</sub>	AlPO <sub>4</sub> /Berlinite, syn	10- 423
3.34 <sub>x</sub>	4.26 <sub>z</sub>	1.82 <sub>z</sub>	1.54 <sub>z</sub>	2.46 <sub>z</sub>	2.28 <sub>z</sub>	1.37 <sub>z</sub>	1.38 <sub>z</sub>	SiO <sub>2</sub> /Quartz, low, syn	33-1161
3.36 <sub>x</sub>	4.23 <sub>x</sub>	3.57 <sub>z</sub>	5.27 <sub>z</sub>	3.72 <sub>z</sub>	4.04 <sub>z</sub>	3.97 <sub>z</sub>	7.19 <sub>z</sub>	C <sub>6</sub> H <sub>6</sub> N <sub>2</sub> O <sub>2</sub> S	30-1944

# Powder diffraction - applications

Qualitative analysis - what IS this stuff?!!



Make d, I list:

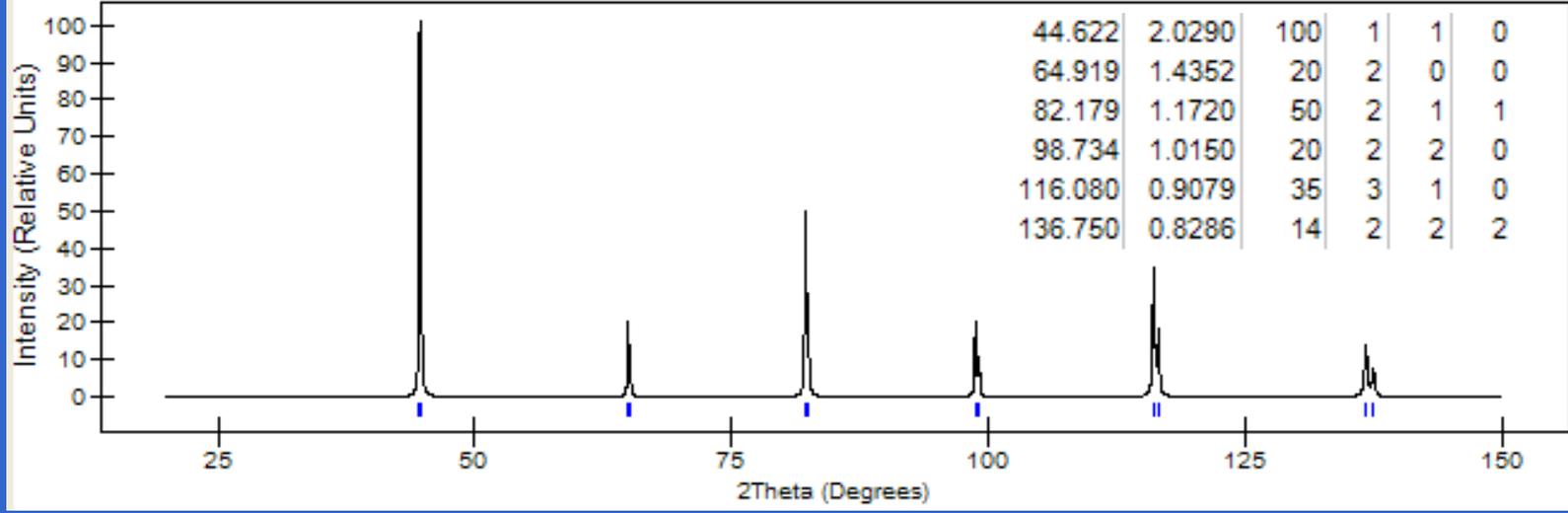
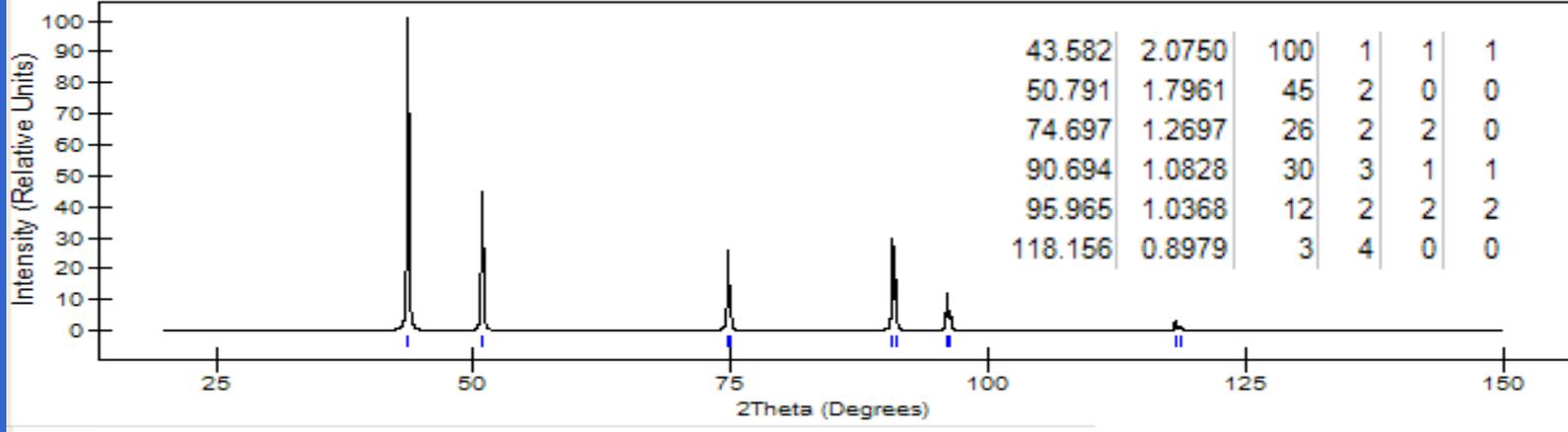
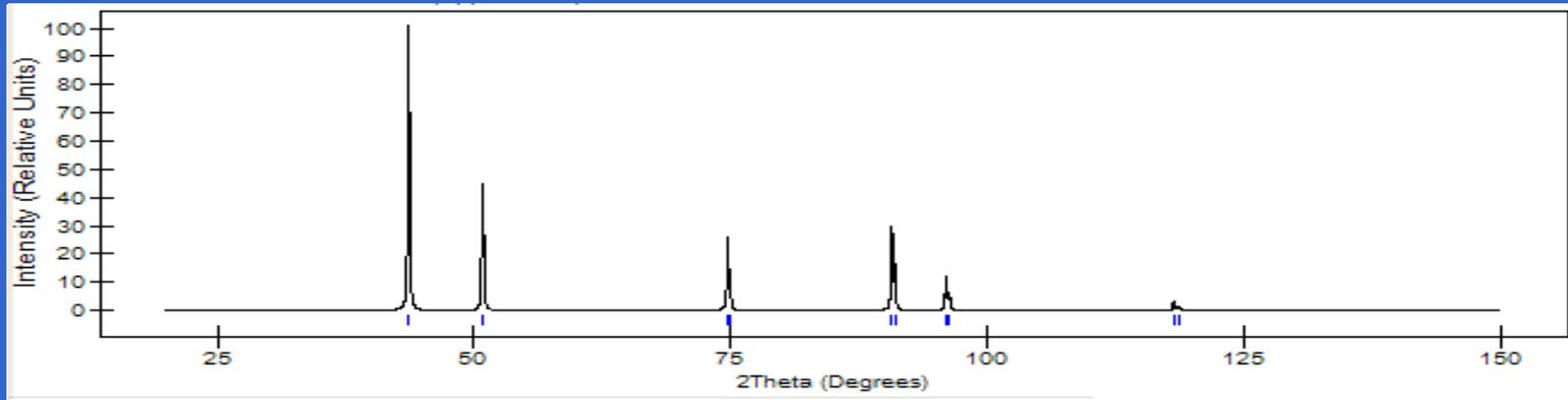
2.080	100	1
2.025	100	2
1.798	45	4
1.433	20	
1.272	26	7
1.170	50	3

1.081	30	6
1.040	12	
1.018	20	8
0.907	35	5
0.898	3	
0.830	14	

Do search/match

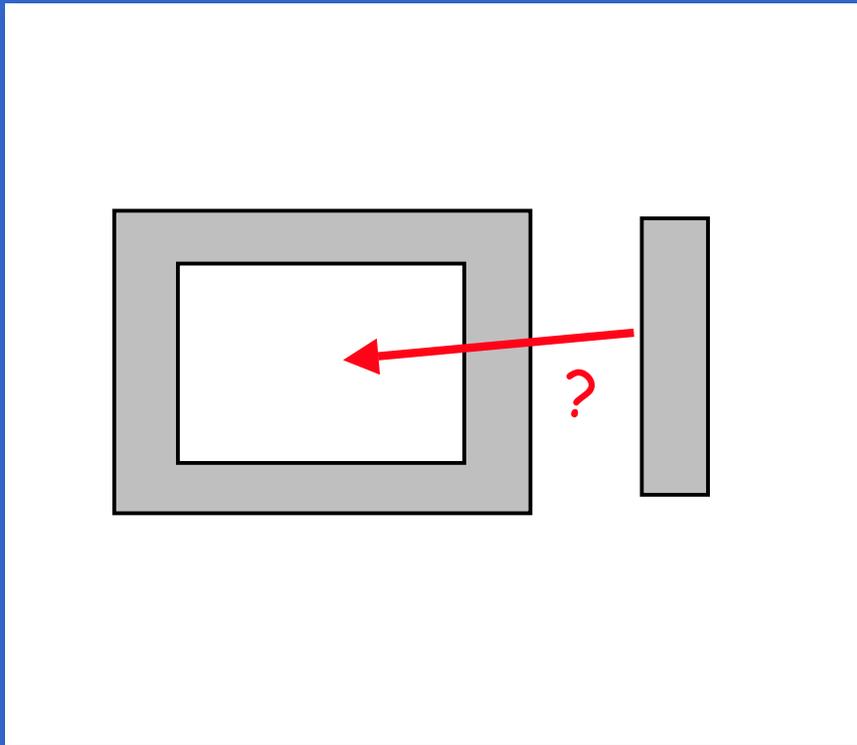
1 4 5 - no match  
 1 4 5 - **MATCH!**  
 1 3 x - no match

2.0750	100	2.0290	100
1.7961	45	1.4352	20
1.2697	26	1.1720	50
1.0828	30	1.0150	20
1.0368	12	0.9079	35
0.8979	3	0.8286	14



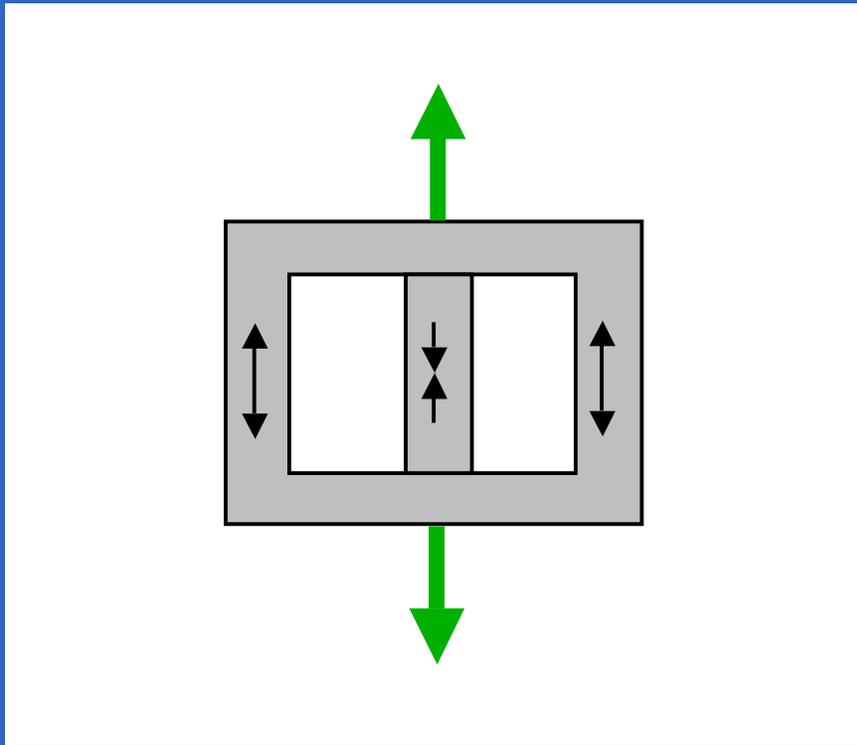
# Powder diffraction - applications

1. Qualitative analysis
2. Quantitative analysis
3. Crystal structure determination
4. Crystallite size
5. Microstrain
6. Residual macrostresses 残余宏观应力



# Powder diffraction - applications

1. Qualitative analysis
2. Quantitative analysis
3. Crystal structure determination
4. Crystallite size
5. Microstrain
6. Residual macrostresses

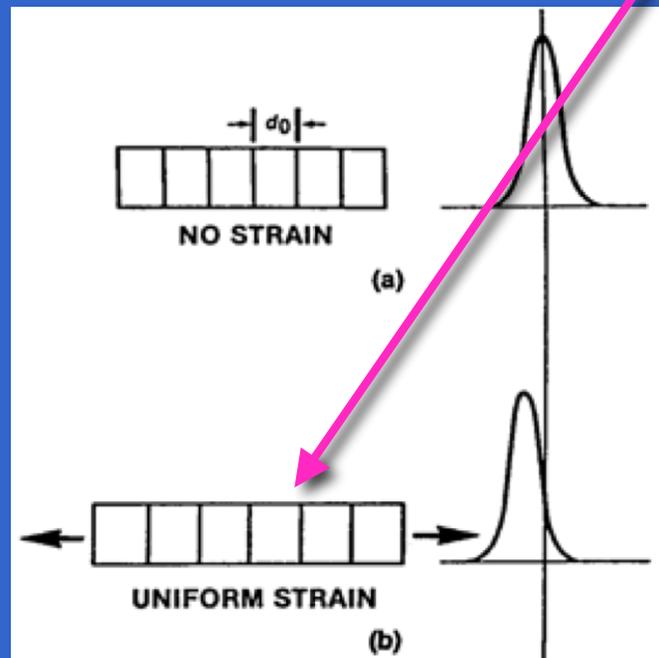


stress = modulus  $\times$  strain  
 $\sigma = E \epsilon$   
(Hooke's law)

# Powder diffraction - applications

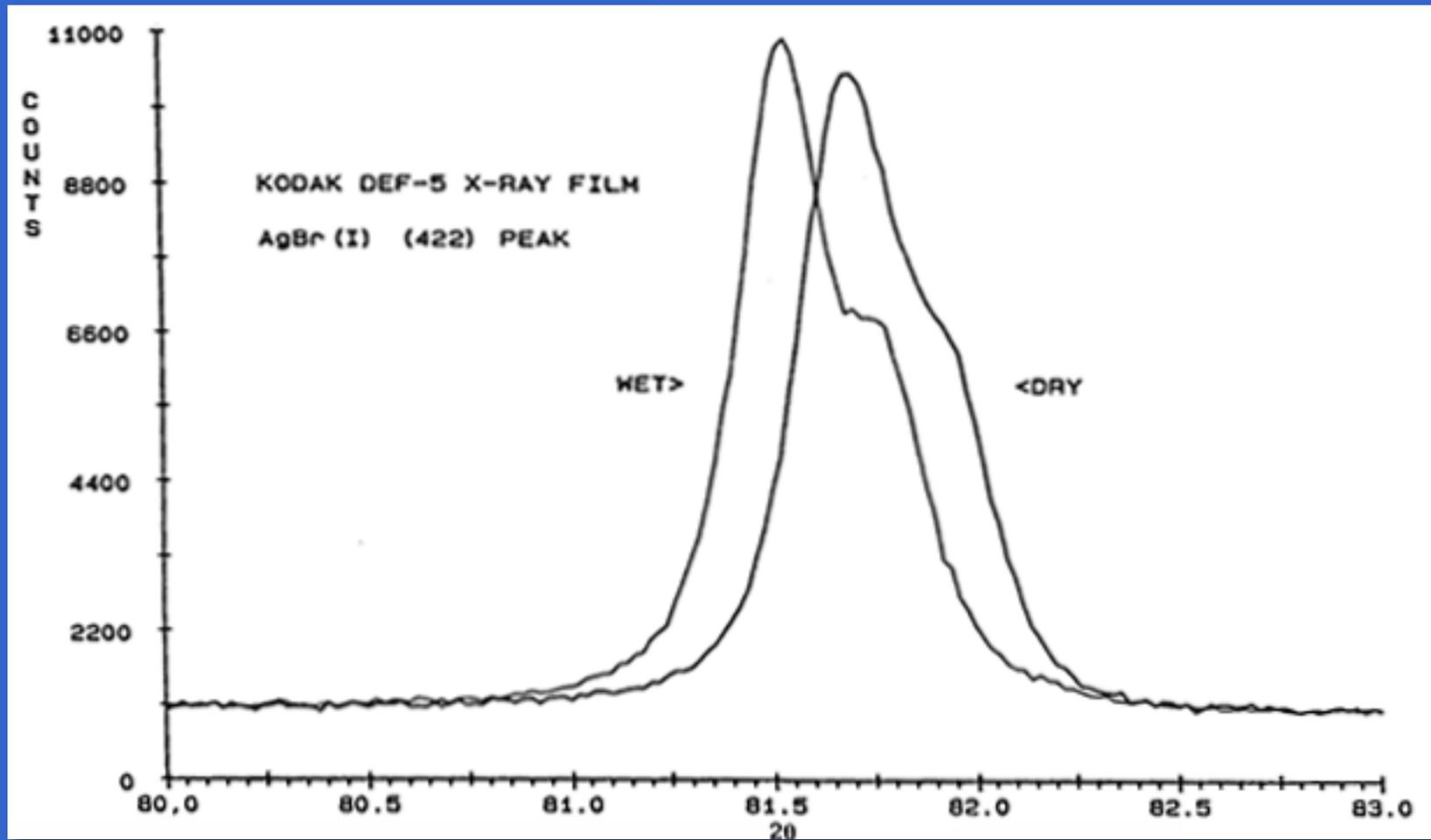
## Residual stress

Reflections shift position (d-values change under stress)



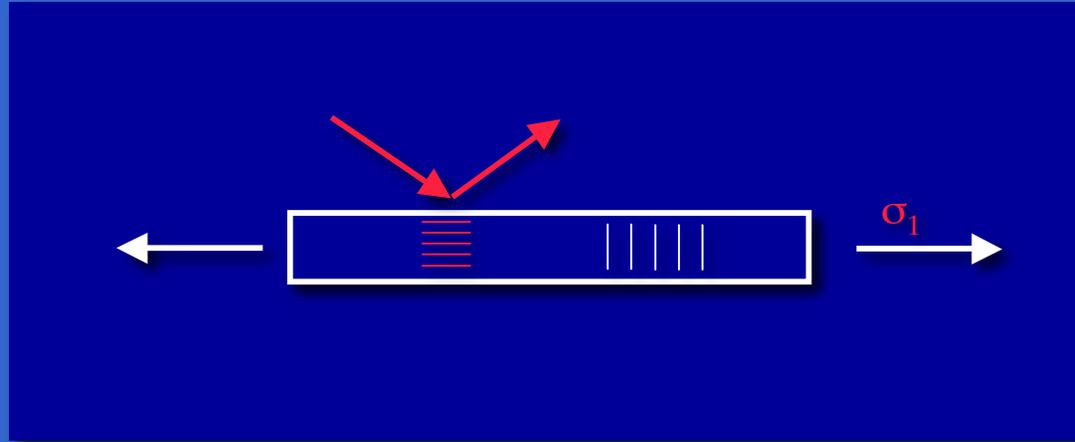
# Powder diffraction - applications

## Residual stresses



# Residual stress

Suppose isotropic 各向同性的 solid loaded in uniaxial 单轴的 tension



Can measure change in interplanar spacing for planes reflecting as above

$$\Delta d = (d - d_0) / d_0 - \text{this is the strain } \varepsilon_3$$

$$\varepsilon_3 = (1/E) (\sigma_3 - \nu(\sigma_1 + \sigma_2)) - \nu \text{ is Poisson's ratio}$$

$$\varepsilon_3 = - (1/E) \nu \sigma_1 - \text{so, can get stress}$$

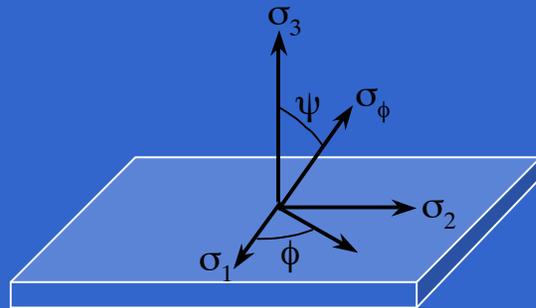
# Residual stress

Biaxial & triaxial stress states more complicated

二轴      三轴

Stress state resolved differently along different directions in sample, so peak shifts change w/ direction

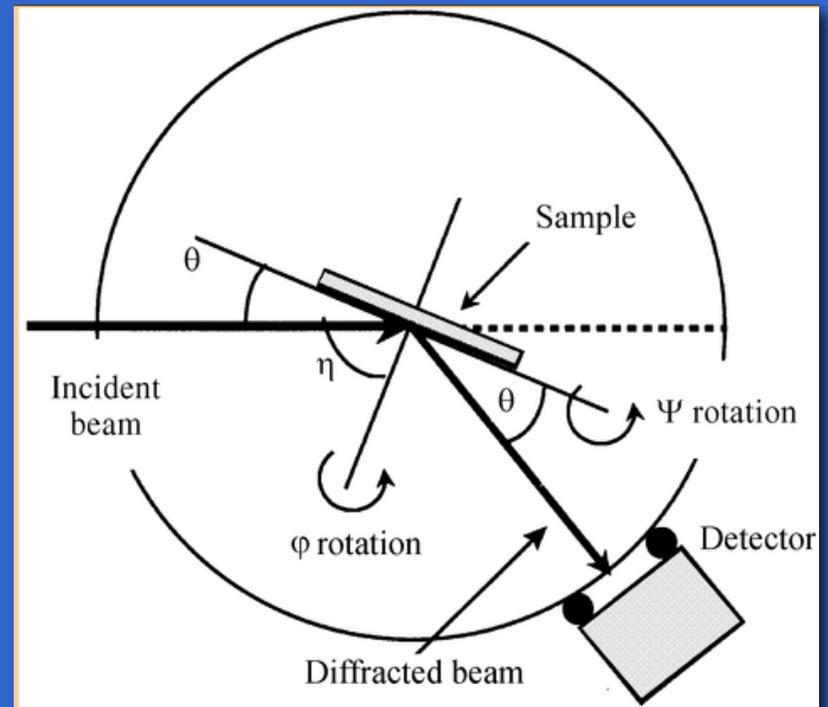
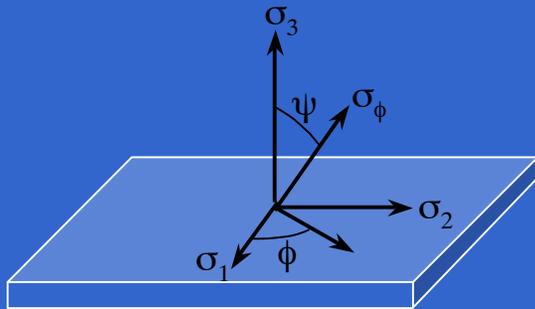
$$\varepsilon_{\phi\psi} = \sigma_{\phi} (1 + \nu)/E \sin^2 \psi - (\nu/E)(\sigma_1 + \sigma_2)$$



# Residual stress

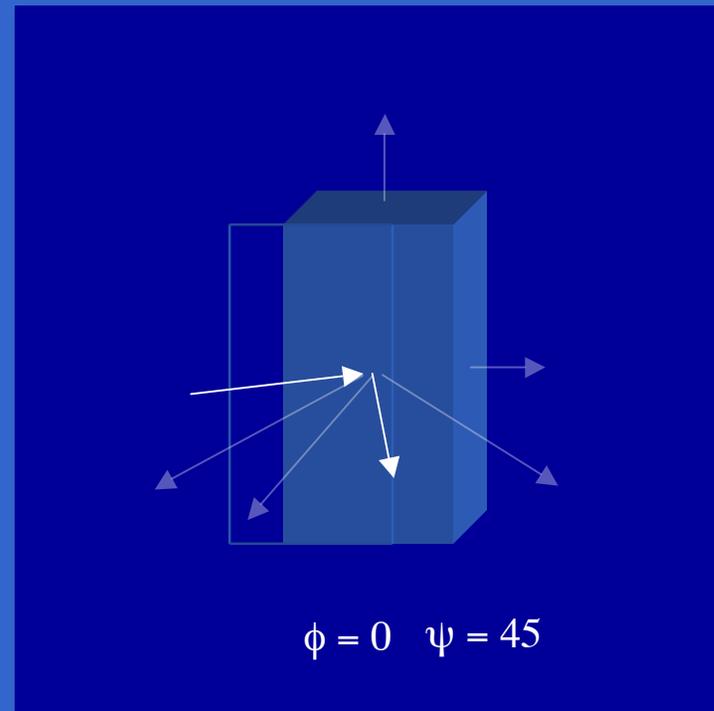
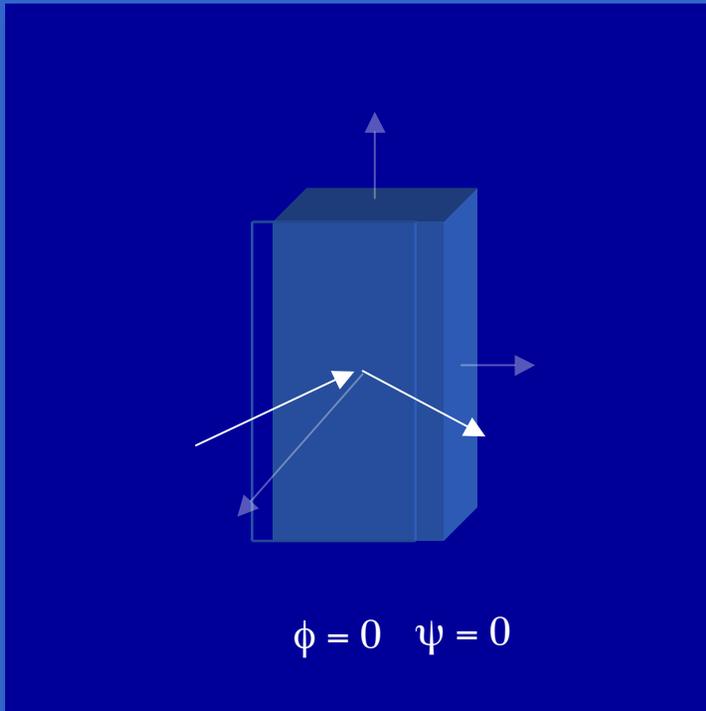
Measurement procedure - classical technique -  $\sin^2 \psi$

Diffractometer must have two more rotations -  $\phi$  and  $\psi$



# Residual stress

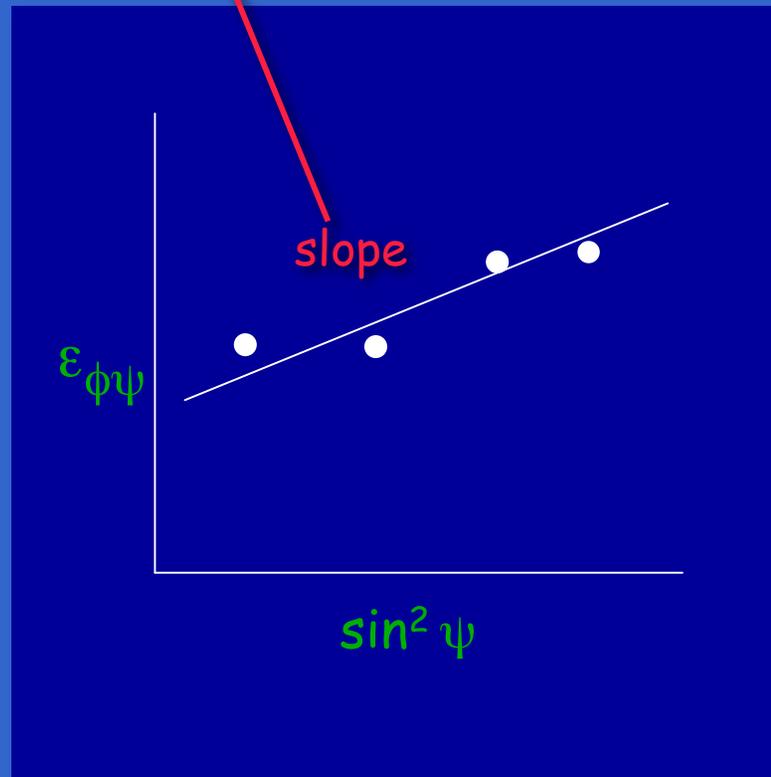
Measurement procedure - classical technique -  $\sin^2 \psi$



# Residual stress

Measurement procedure - classical technique -  $\sin^2 \psi$

$$\varepsilon_{\phi\psi} = \sigma_{\phi} (1 + \nu)/E \sin^2 \psi - (\nu/E)(\sigma_1 + \sigma_2)$$



Need  $E, \nu$  -  
measure or  
calculate

# Powder diffraction - applications

1. Qualitative analysis
2. Quantitative analysis
3. Crystal structure determination
4. Crystallite size
5. Microstrain
6. Residual macrostresses
7. Thin films

# Powder diffraction - applications

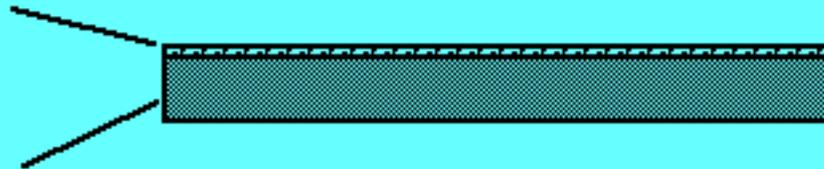
## Thin films

### Epitaxy in polycrystalline thin films

外延

thin film

substrate

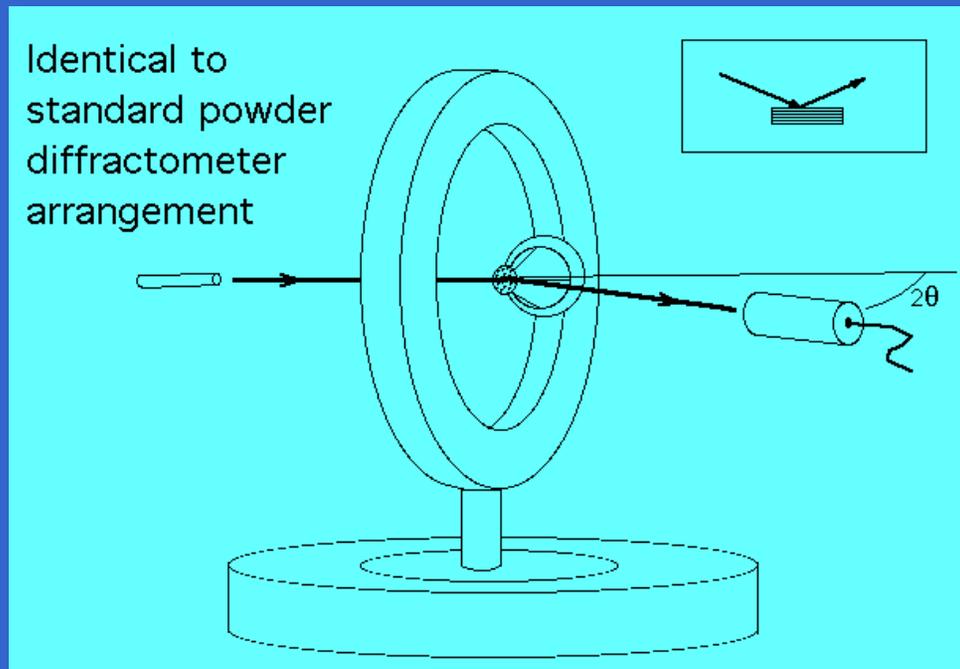


# Powder diffraction - applications

## Thin films

First, suppose we have a thin  
single crystal plate

Mount in 4-circle diffractometer

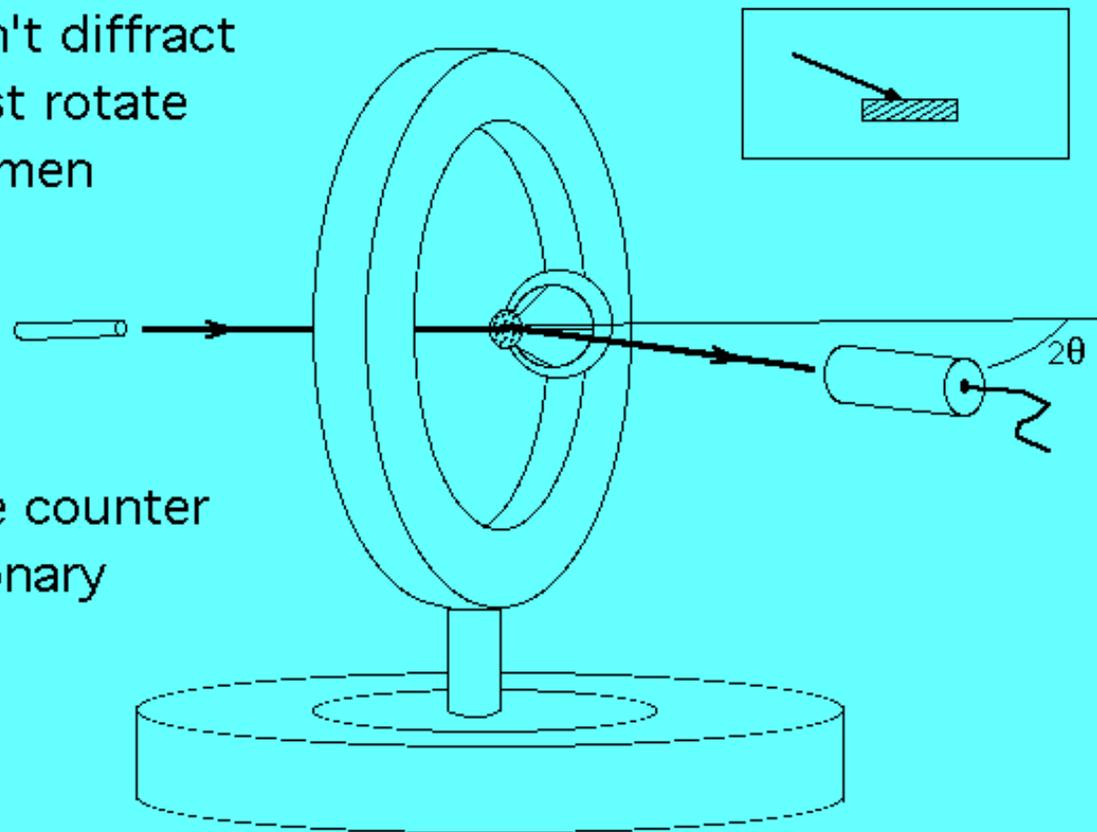


# Powder diffraction - applications

## Thin films

This set of planes (same  $d$ )  
doesn't diffract  
- must rotate  
specimen

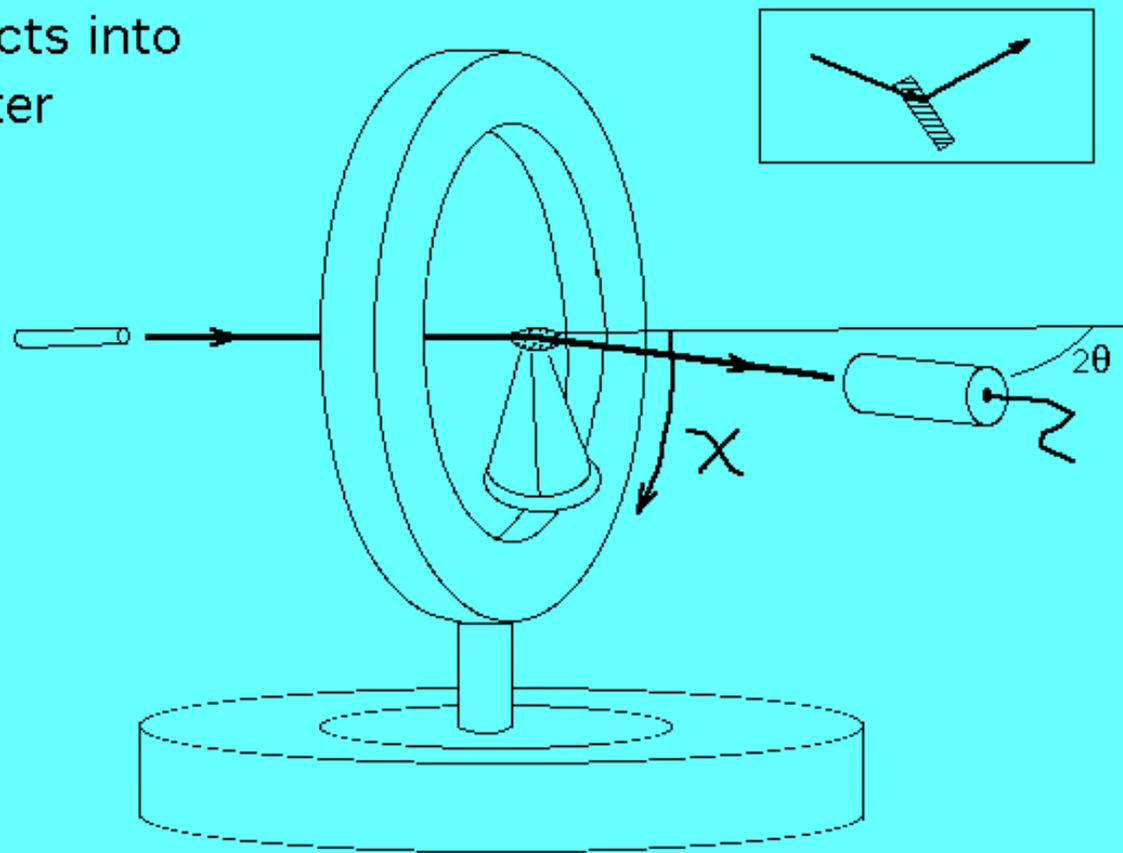
Leave counter  
stationary



# Powder diffraction - applications

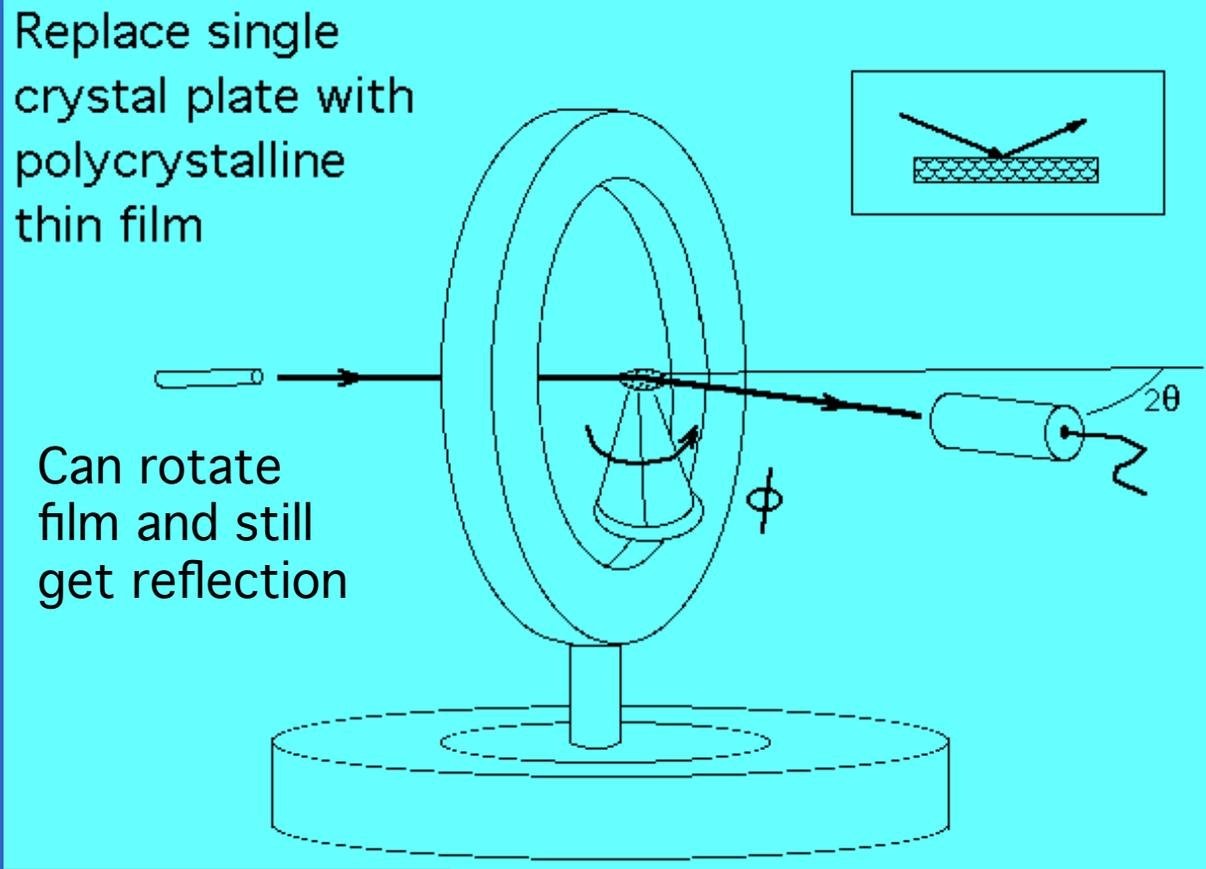
## Thin films

Now this set of planes  
diffracts into  
counter



# Powder diffraction - applications

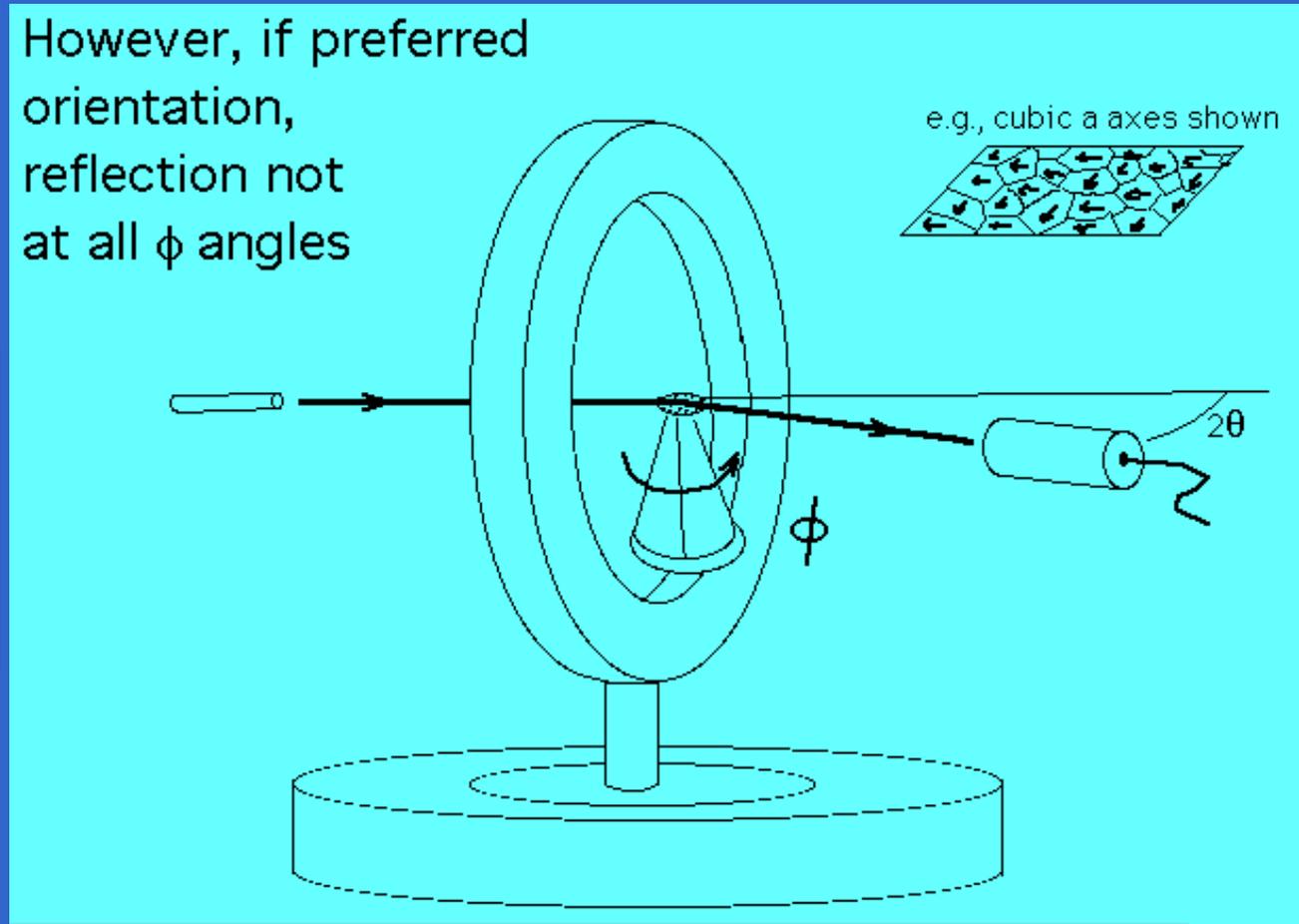
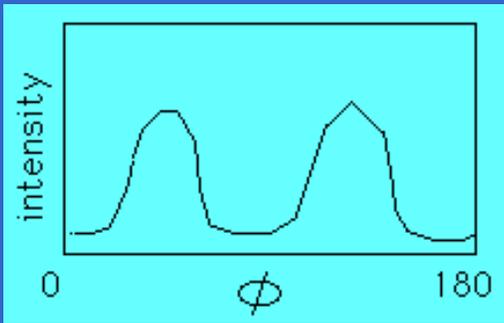
## Thin films



# Powder diffraction - applications

## Thin films

However, if preferred orientation, reflection not at all  $\phi$  angles



# Powder diffraction - applications

1. Qualitative analysis
2. Quantitative analysis
3. Crystal structure determination
4. Crystallite size
5. Microstrain
6. Residual macrostresses
7. Thin films
8. Crystallinity
9. Studies of phase diagrams
  - A. Phase boundaries
  - B. Solid solutions
  - C. Phase transitions
10. Studies of chemical reactions and identification of reaction products.
11. Determination of grain sizes.
12. Preferred orientation in polycrystalline bodies.
13. Many more