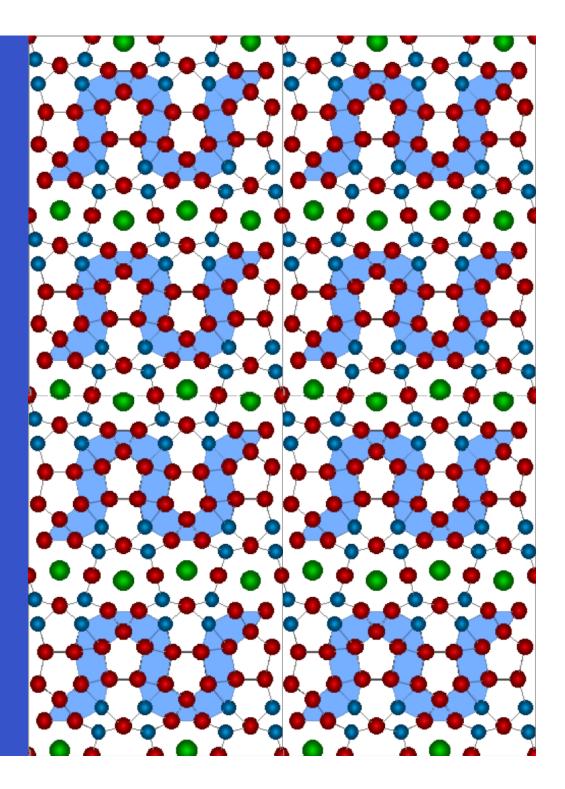
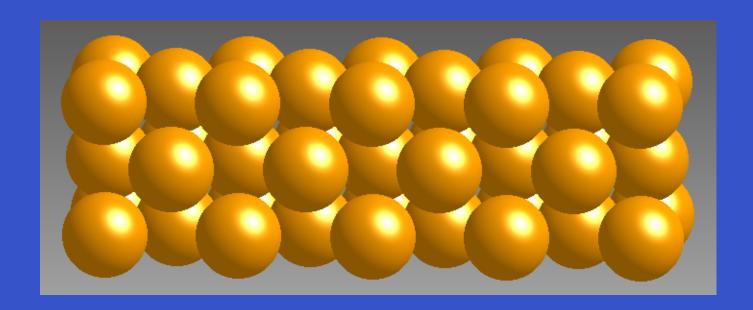
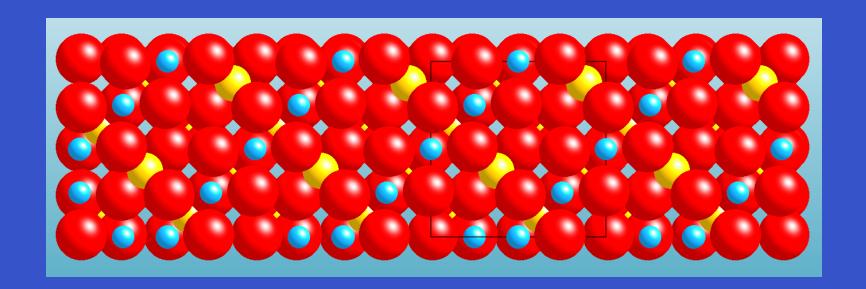
crystallography (晶体学)



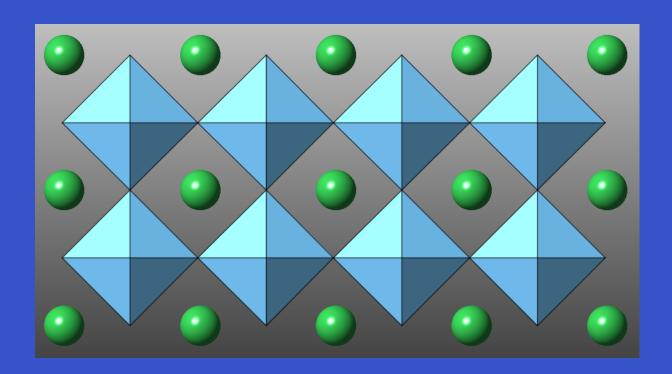
Crystalline means atom arrangement is periodic (周期性的) - repeats throughout space



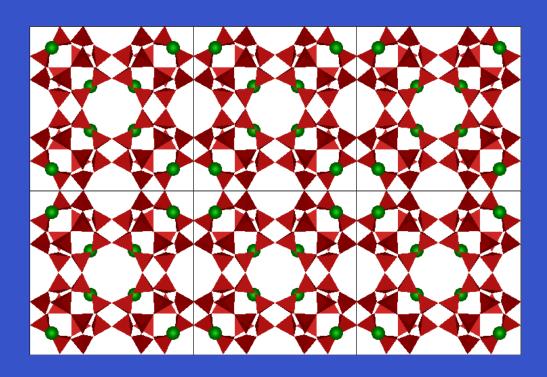
metals



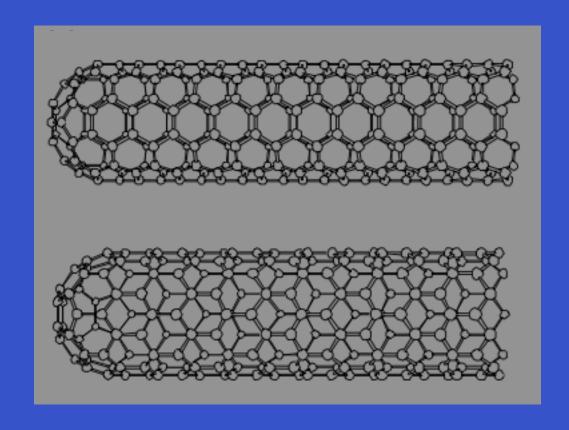
spinels (ferrites - 铁电的)



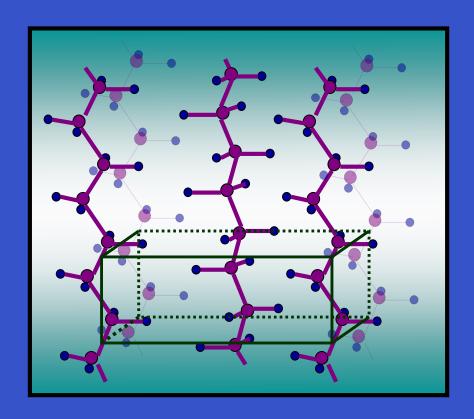
BaTiO<sub>3</sub> (钡钛氧<sub>3</sub>)



zeolite (沸石)



carbon nanotubes (纳米管) (CNTs)



polyethylene (聚乙烯) polymer

# This type of structure is called <a href="mailto:crystal structure">crystal structure</a> (晶体结构)

In crystals, atom groups (unit cells) repeated to form a solid material

The study of this repetition in crystals is called crystallography

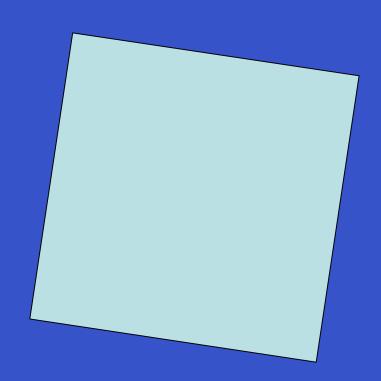
#### Repetition = symmetry (对称性)

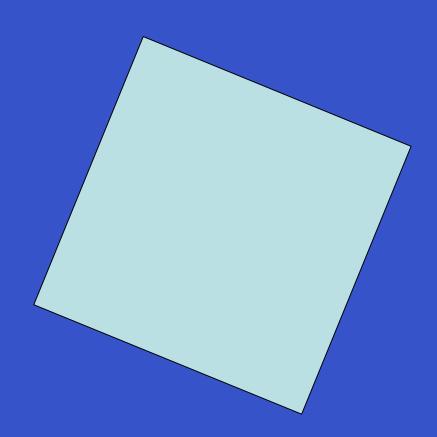
Types of repetition:

rotation (旋转) translation (平移)

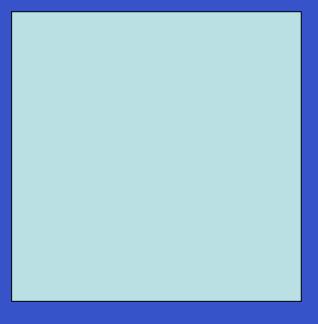
#### Rotation

What is rotational symmetry?





#### Was it?



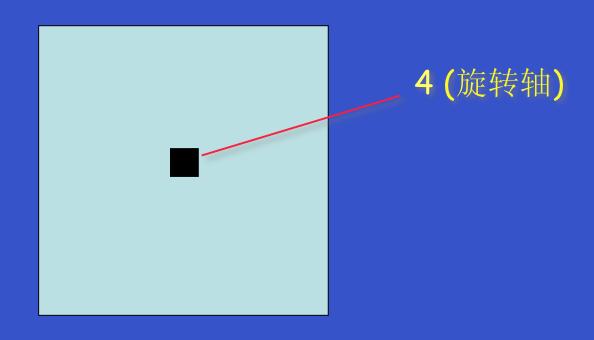
### The object is obviously symmetric...it has symmetry

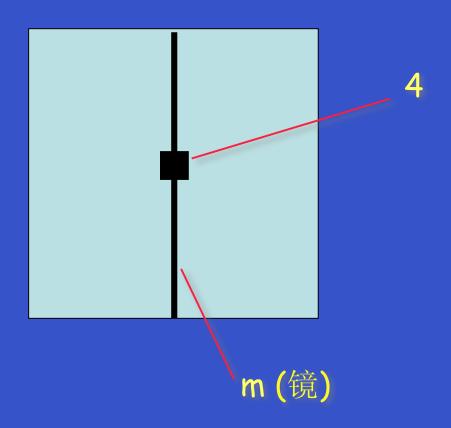
The object is obviously symmetric...it has symmetry

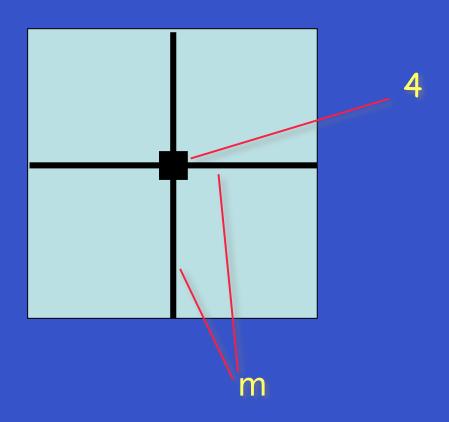
Can be rotated 90° w/o detection

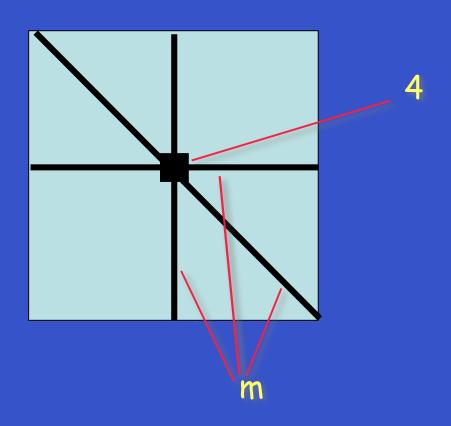
## .....so symmetry is really doing nothing

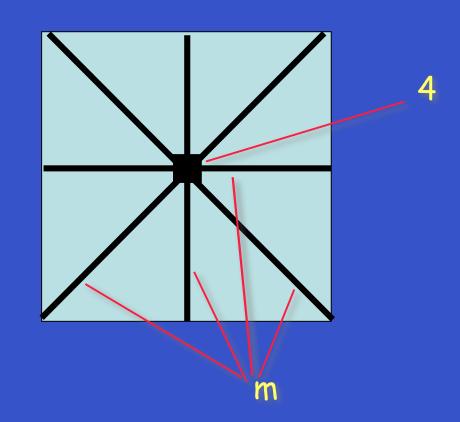
Symmetry is doing nothing - or at least doing something so that it looks like nothing was done!





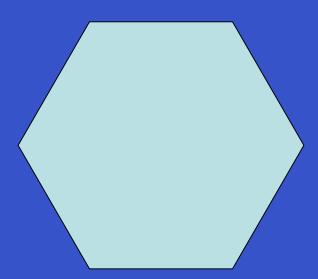




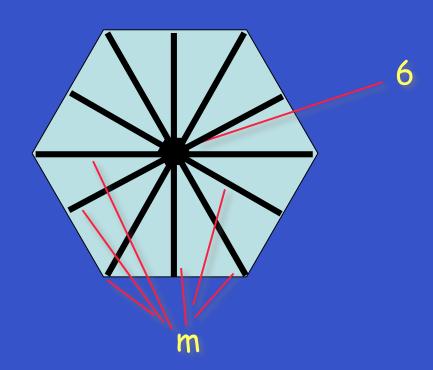


4mm (点群)

#### Another example:



#### Another example:

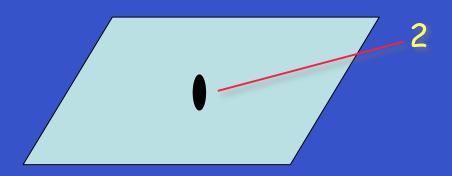


6mm

#### And another:



#### And another:



2

#### What about translation?

Same as rotation

#### What about translation?

Same as rotation

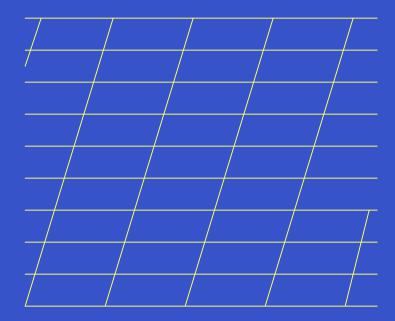
Ex: one dimensional array of points



Translations are restricted to only certain values to get symmetry (periodicity)

#### 2D translations

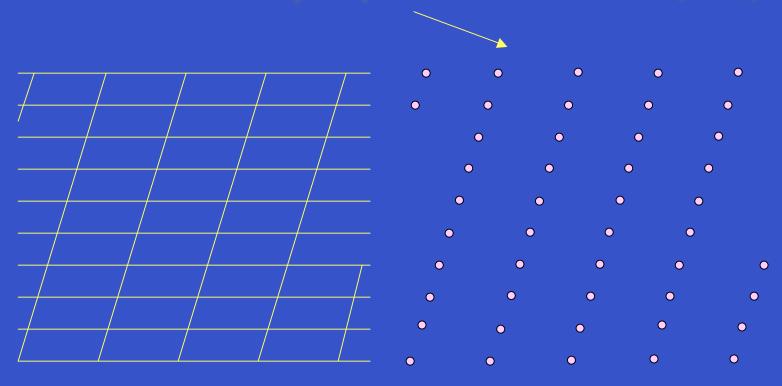
#### Lots of common examples



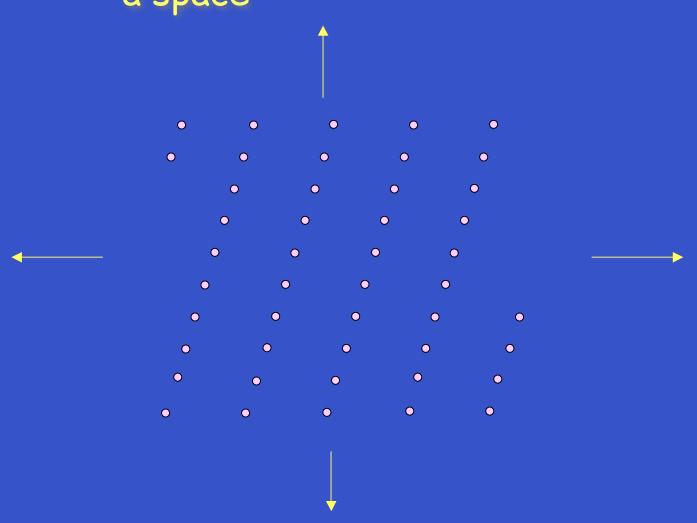
#### Each block is represented by a point



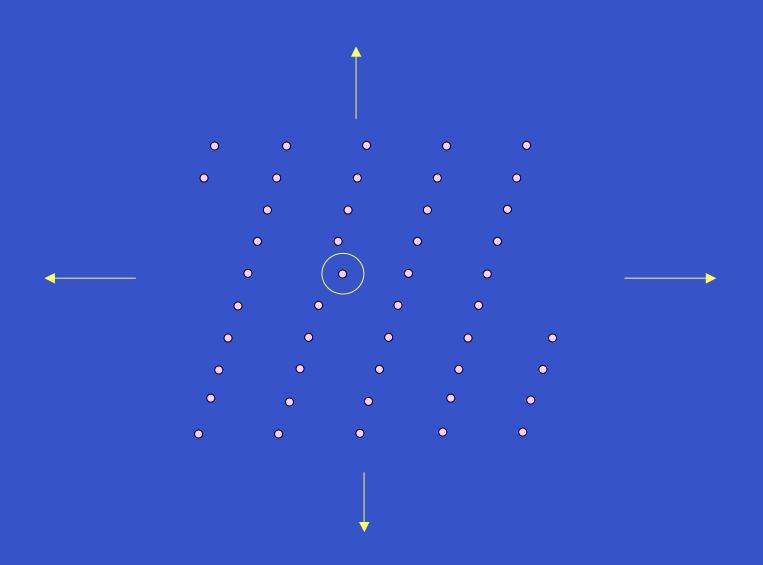
#### This array of points is a LATTICE (晶格)



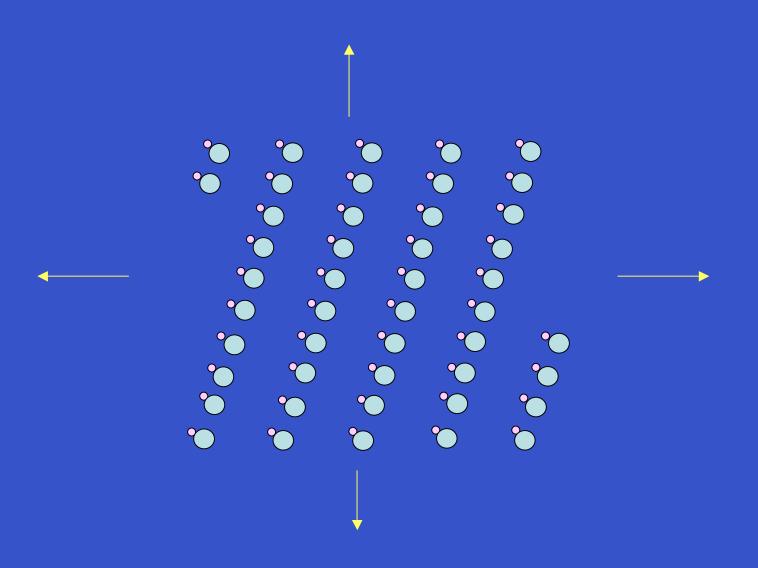
# Lattice - infinite (无限的), perfectly periodic (周期性的) array of points in a space



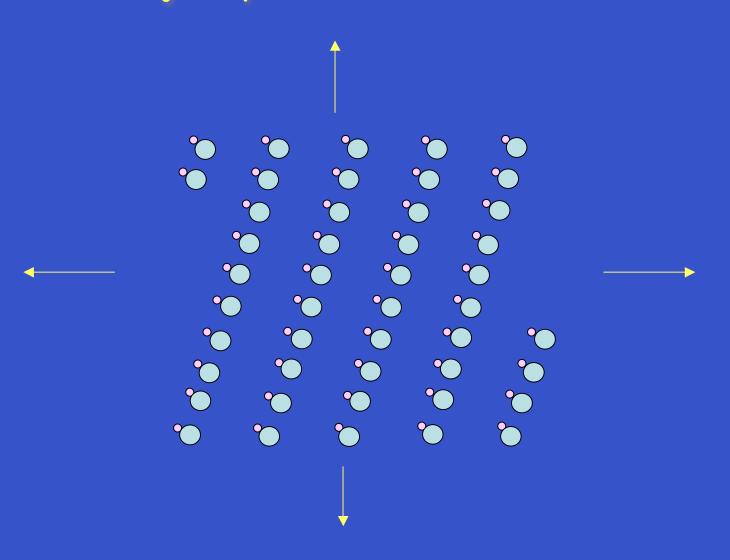
## Not lattice:



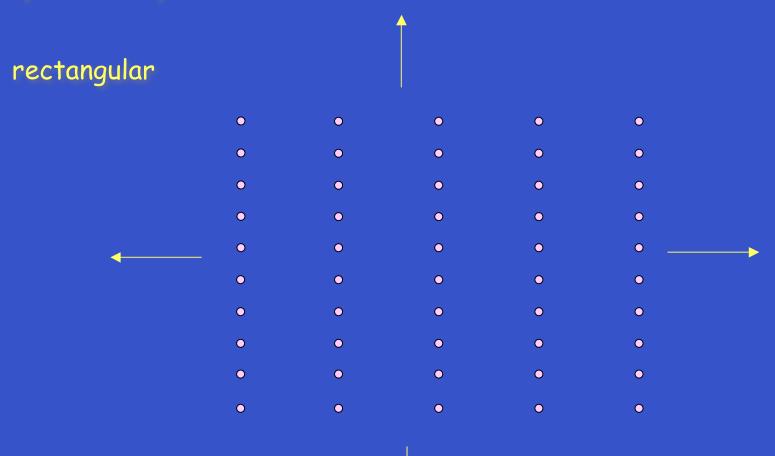
### Not lattice:



## Not lattice - ....some kind of STRUCTURE because not just points



# Another type of lattice - with different symmetry



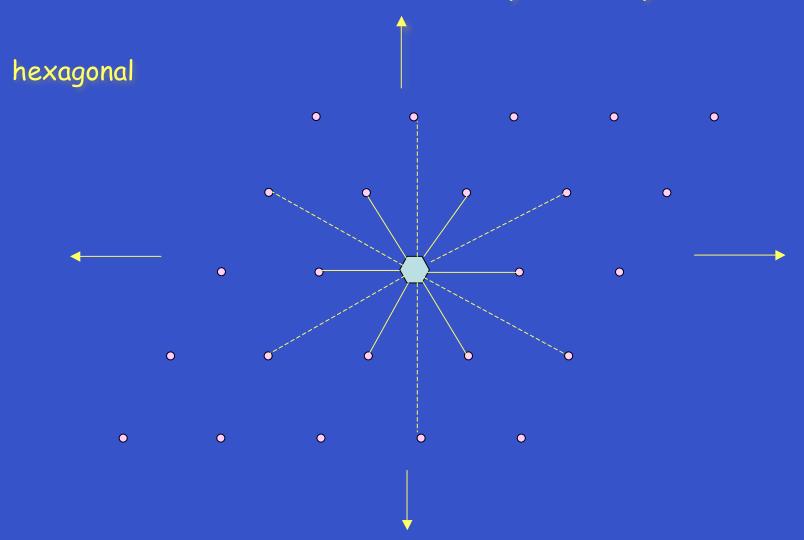
# Another type of lattice - with different symmetry

square

# Another type of lattice - with different symmetry

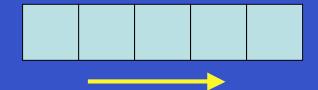
```
hexagonal (六角)
               0
```

## Back to rotation -This lattice exhibits 6-fold symmetry

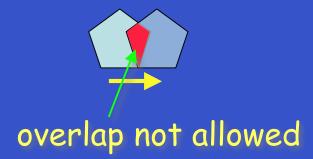


## Periodicity and rotational symmetry

What types of rotational symmetry allowed?



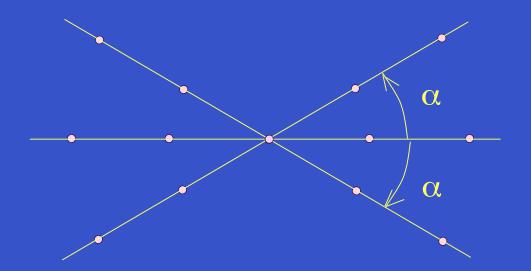
object with 4-fold symmetry translates OK



object with 5-fold symmetry doesn't translate

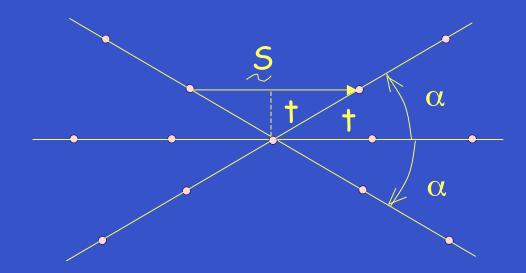
## Periodicity and rotational symmetry

Suppose periodic row of points is rotated through  $\pm \alpha$ :

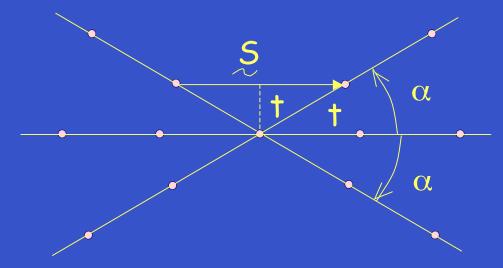


## Periodicity and rotational symmetry

To maintain periodicity:



vector S = an integer x basis translation <math>t



vector S = an integer x basis translation <math>t $t \cos \alpha = S/2 = mt/2$ 

m	$\cos \alpha$	$\alpha$	axis
2	1	0 2π	1
1	1/2	$\pi/3 5\pi/3$	6
0	0	$\pi/2$ $3\pi/2$	4
-1	-1/2	$2\pi/3 \ 4\pi/3$	3
-2	-1	-π π	2

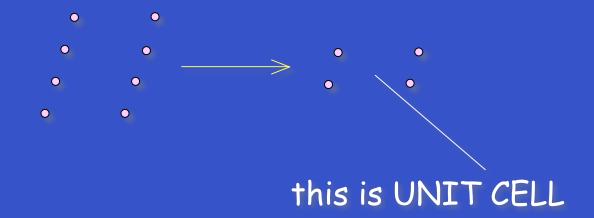
m	cos a	α	axis
2	1	0 2π	
	1		
1	1/2	π/3 5π/	3 6
0	0	$\pi/2$ $3\pi/$	2 4
-1	-1/2	$2\pi/3 \ 4\pi/3$	3 3
-2	-1	- π π	2

Only rotation axes consistent with lattice periodicity in 2-D or 3-D

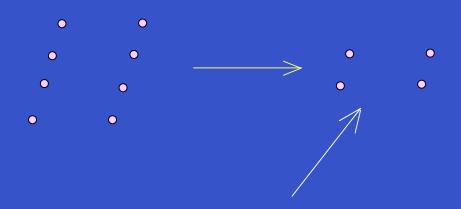
#### We abstracted points from the shape:



#### Now we abstract further:



#### Now we abstract further:



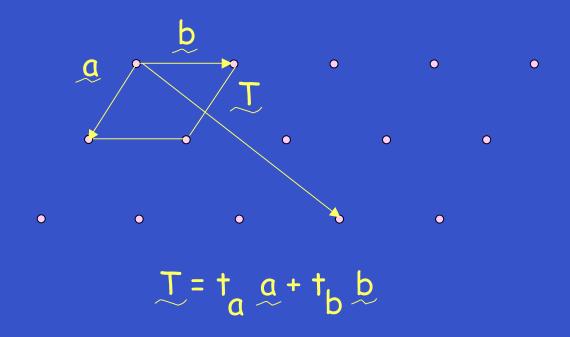
#### This is a UNIT CELL

Represented by two lengths and an angle



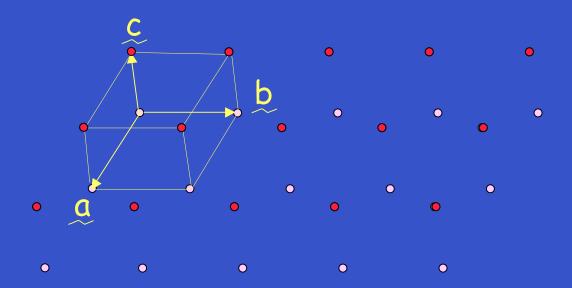
.....or, alternatively, by two vectors

## Basis vectors and unit cells



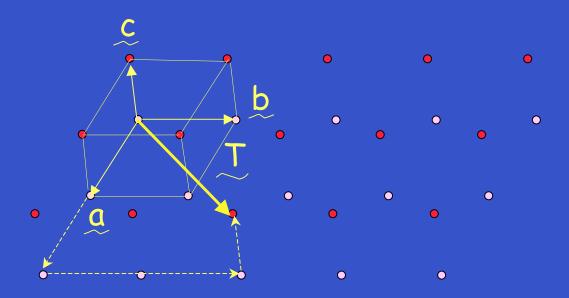
a and b are the basis vectors for the lattice

## In 3-D:



a, b, and c are the basis vectors for the lattice

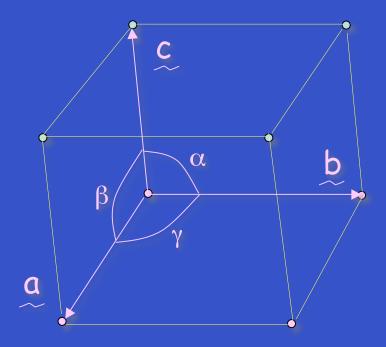
#### In 3-D:



 $T = t_a \underline{a} + t_b \underline{b} + t_c \underline{c} \longrightarrow [221]$  direction

a, b, and c are the basis vectors for the lattice

## Lattice parameters (晶格参数):



need 3 lengths - |a|, |b|, |c| & 3 angles -  $\alpha$ ,  $\beta$ ,  $\gamma$  to get cell shape & size

# The many thousands of lattices classified into crystal systems (晶系)

System	Interaxial Angles	Axes
Triclinic Monoclinic Orthorhombic Tetragonal Cubic Hexagonal Trigonal	$\alpha \neq \beta \neq \gamma \neq 90^{\circ}$ $\alpha = \gamma = 90^{\circ} \neq \beta$ $\alpha = \beta = \gamma = 90^{\circ}$ $\alpha = \beta = \gamma = 90^{\circ}$ $\alpha = \beta = \gamma = 90^{\circ}$ $\alpha = \beta = 90^{\circ}, \gamma = 120^{\circ}$ $\alpha = \beta = 90^{\circ}, \gamma = 120^{\circ}$	a ≠ b ≠ c a ≠ b ≠ c a ≠ b ≠ c a = b ≠ c a = b ≠ c a = b ≠ c a = b ≠ c

# The many thousands of lattices classified into crystal systems

System	Minimum symmetry
Triclinic	$1$ or $\overline{1}$
Monoclinic	$2 \text{ or } \overline{2}$
Orthorhombic	three $2s$ or $2s$
Tetragonal	$4 \text{ or } \overline{4}$
Cubic	four 3s or $\overline{3}$ s
Hexagonal	$6 \text{ or } \overline{6}$
Trigonal	$3 \text{ or } \overline{3}$

# For given lattice, infinite number of unit cells possible:

